## The atom in electric field

- The electric field in the direction of the Oz axis
- The interaction with the electric field

$$H_E = \mathcal{E}\sum_i z_i$$

- We assume, it is stronger than the spin-orbit interaction (valid for E>10<sup>5</sup> V/m
- For the hydrogen atom the energy levels are degenerate (except the ground state)
- For the ground state the first-order perturbation correction is

$$E_E = \mathcal{E}\langle 1s|z|1s\rangle$$

• Because z is an uneven function, this integral is 0

• More general reasoning:

$$z = r\cos\theta = \frac{2}{\sqrt{3}}Y_{10}(\theta)$$

• The orbital part of the integral:

$$\int d\hat{\mathbf{r}} Y_{00}^*(\hat{\mathbf{r}}) Y_{10}(\hat{\mathbf{r}}) Y_{00}(\hat{\mathbf{r}}) = 0.$$

• Because

$$\int Y_{l_a m_a}^*(\hat{\mathbf{r}}) Y_{l_b m_b}(\hat{\mathbf{r}}) Y_{l_c m_c}(\hat{\mathbf{r}}) d\hat{\mathbf{r}} = \sqrt{\frac{(2l_b+1)(2l_c+1)}{4\pi(2l_a+1)}} C_{l_b 0 l_c 0}^{l_a 0} C_{l_b m_b l_c m_c}^{l_a m_a}$$

 The first-order perturbation correction for the ground state is 0, there is no linear Stark effect

- The excited states are n<sup>2</sup>-fold degenerate in respect with I and m<sub>I</sub>
- We apply the perturbation method for n=2

• Here

$$H_{ss} = \langle 2s|H_E|2s \rangle$$
  

$$H_{spm_l} - \langle 2s|H_E|2p_{m_l} \rangle$$
  

$$H_{pm_lpm'_l} = \langle 2p_{m_l}|H_E|2p_{m'_l} \rangle.$$

 All matrix elements are zero, except that between 2s and 2p<sub>0</sub>

$$H_{sp0} = \langle 2s|H_E|2p_0 \rangle = \langle 2p_0|H_E|2s \rangle$$
  
$$= \mathcal{E}\frac{2}{\sqrt{3}} \int d\hat{\mathbf{r}} Y_{00}^*(\hat{\mathbf{r}}) Y_{10}(\hat{\mathbf{r}}) \int_0^\infty dr r^3 R_{2s}(r) R_{2p}(r)$$
  
$$= -\frac{3}{Z} \mathcal{E}, \qquad ($$

The equation becomes

$$\begin{vmatrix} -E_E & H_{sp0} & 0 & 0 \\ H_{sp0} & -E_E & 0 & 0 \\ 0 & 0 & -E_E & 0 \\ 0 & 0 & 0 & -E_E \end{vmatrix} = 0$$

• The solutions are

$$E_E(\pm 1) = 0$$
 for m<sub>l</sub>=+1 and -1 
$$E_E(0)_{1,2} = \pm H_{sp0} = \mp \frac{3}{Z} \mathcal{E}$$
 for m<sub>l</sub>=0

#### The wavefunctions are

$$\psi_1 = \frac{1}{\sqrt{2}}(2s - 2p_0)$$

#### for the higher energy level, and

$$\psi_2 = \frac{1}{\sqrt{2}}(2s + 2p_0)$$

#### for the lower energy level.

 For n=2 we have the linear Stark effect (proportional with E), because this degenerate level has not a specific parity

#### Energy term diagram for the linear Stark effect

n=3



## Higher energy levels

#### Stark effect in hydrogen



Taking into account the spin-orbit splitting

Stark effect of Hydrogen for n=3 and n=2



 For the ground state we calculate the second-order perturbation correction

$$E_{E}^{(2)} = \sum_{n \neq 1, l, m} \frac{|\langle \psi_{nlm} | H_E | 1s \rangle|^2}{E_1 - E_n}$$
$$= \mathcal{E}^2 \sum_{n \neq 1, l, m} \frac{|\langle \psi_{nlm} | z | 1s \rangle|^2}{E_1 - E_n}$$

 Replacing En by E2 we obtain the upper limit (in absolute value) of the correction

$$E_E^{(2)} > \mathcal{E}^2 \frac{1}{E_1 - E_2} \sum_{n \neq 1, l, m} |\langle \psi_{nlm} | z | 1s \rangle|^2$$

Taking into account that

$$\langle 1s|z|1s\rangle = 0$$

$$\sum_{n \neq 1, l, m} |\langle \psi_{nlm} | z | 1s \rangle|^2 = \sum_{n, l, m} |\langle \psi_{nlm} | z | 1s \rangle|^2$$
$$= \sum_{n, l, m} \langle 1s | z | \psi_{nlm} \rangle \langle \psi_{nlm} | z | 1s \rangle$$
$$= \langle 1s | z^2 | 1s \rangle,$$

Where we have used the closure relationship

$$\sum_{n,l,m} |\psi_{nlm}\rangle \langle \psi_{nlm}| = 1$$

The matrix element can be calculated analytically

$$\langle 1s|z^2|1s\rangle = \frac{1}{Z^2}$$

• With 
$$E_1 = -Z^2/2$$
 and  $E_2 = -Z^2/8$ 

we obtain

$$E_E^{(2)} > -\frac{8}{3} \frac{\mathcal{E}^2}{Z^4}$$

#### The exact solution leads to

$$E_E^{(2)} = -2,25\frac{\mathcal{E}^2}{Z^4}$$

#### This correction is the quadratic Stark effect

### **Multielectron atoms**

• We introduce

$$D_z = -\sum_i z_i$$

the z component of the electric dipole .

The

$$H_E = \mathcal{E}\sum_i z_i$$

perturbation leads to

$$E_E^{(1)} = -\mathcal{E}\langle kLSJM_J | D_z | kLSJM_J \rangle$$

The unperturbed energy levels are not degenerated in respect with L, the  $|kLSJM_J\rangle$  states have a certain parity.

 Because the dipole operator has odd parity, all the matrix elements will be zero.

$$E_E^{(1)} = 0,$$

- For multielectron atoms there is not linear Stark effect
- The quadratic Stark effect (second-order perturbation correction):

$$E_E^{(2)} = \mathcal{E}^2 \sum_{k'L'S'J'M_J'} \frac{|\langle kLSJM_J | D_z | k'L'S'J'M_J' \rangle|^2}{E_{kLSJ} - E_{k'L'S'J'}}$$

After some calculations one obtains

$$E_E^{(2)} = \mathcal{E}^2(a + bM_J^2)$$

The degeneracy is only partly removed

# Stark effect splitting of the helium transition at 438.8 nm.



#### The multielectron atoms has no electric dipole momentum, and this is the reason why they show no linear Stark effect.

- The quadratic Stark effect may be interpreted as the induction of the dipole momentum by the external electric field, and the interaction of the induced momentum with this field.
- The hydrogen atom behaves, as if it would have electric dipole momentum.