

Recent development of rare earth lean permanent magnets

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With the financial supported of

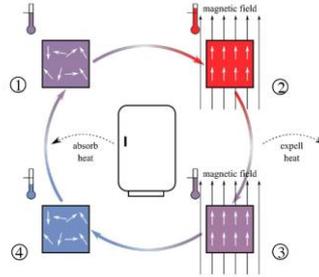
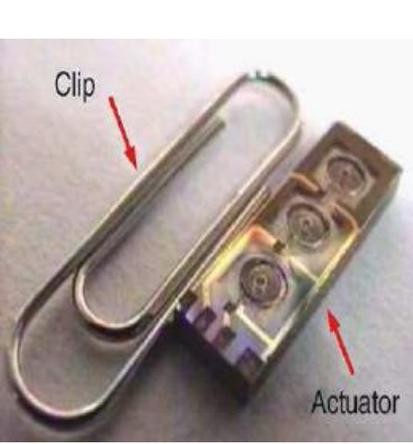
Romanian Ministry of Education and Research, grant PN-II-ID-PCE-2012-4-0470

Outline

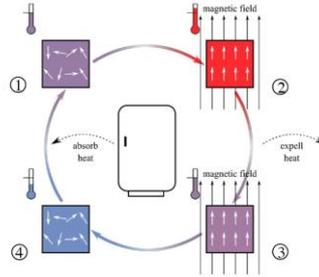
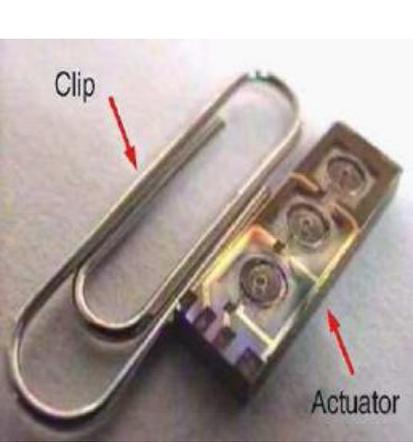
- **Introduction**
- **MnBi magnetic phase**
- **MnAl magnetic phase**
- **Nanocomposites magnets=Spring magnets**
- **Conclusions**

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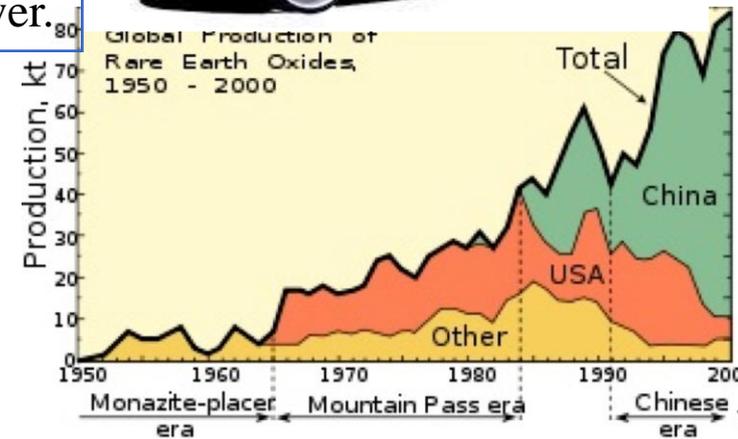
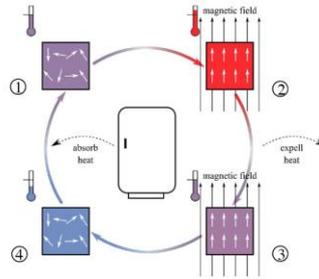
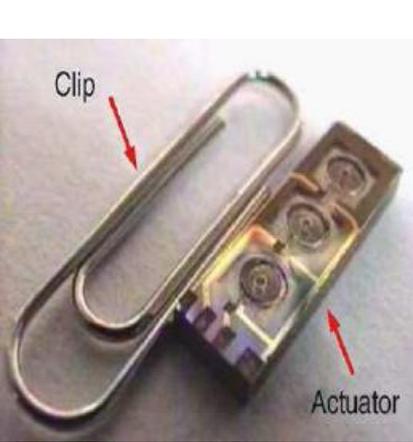
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➤ magnetic materials are *critical* components in many devices and for advanced technologies.



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- high performance magnet (HPM)/wind generator **1000-1600 kg/MW**.
- motors and generators: **2 kg** HPM/hybrid electric vehicle-**20 million** vehicles by 2018.
- Magnetocaloric applications **4 kg HPM/kW** cooling power.

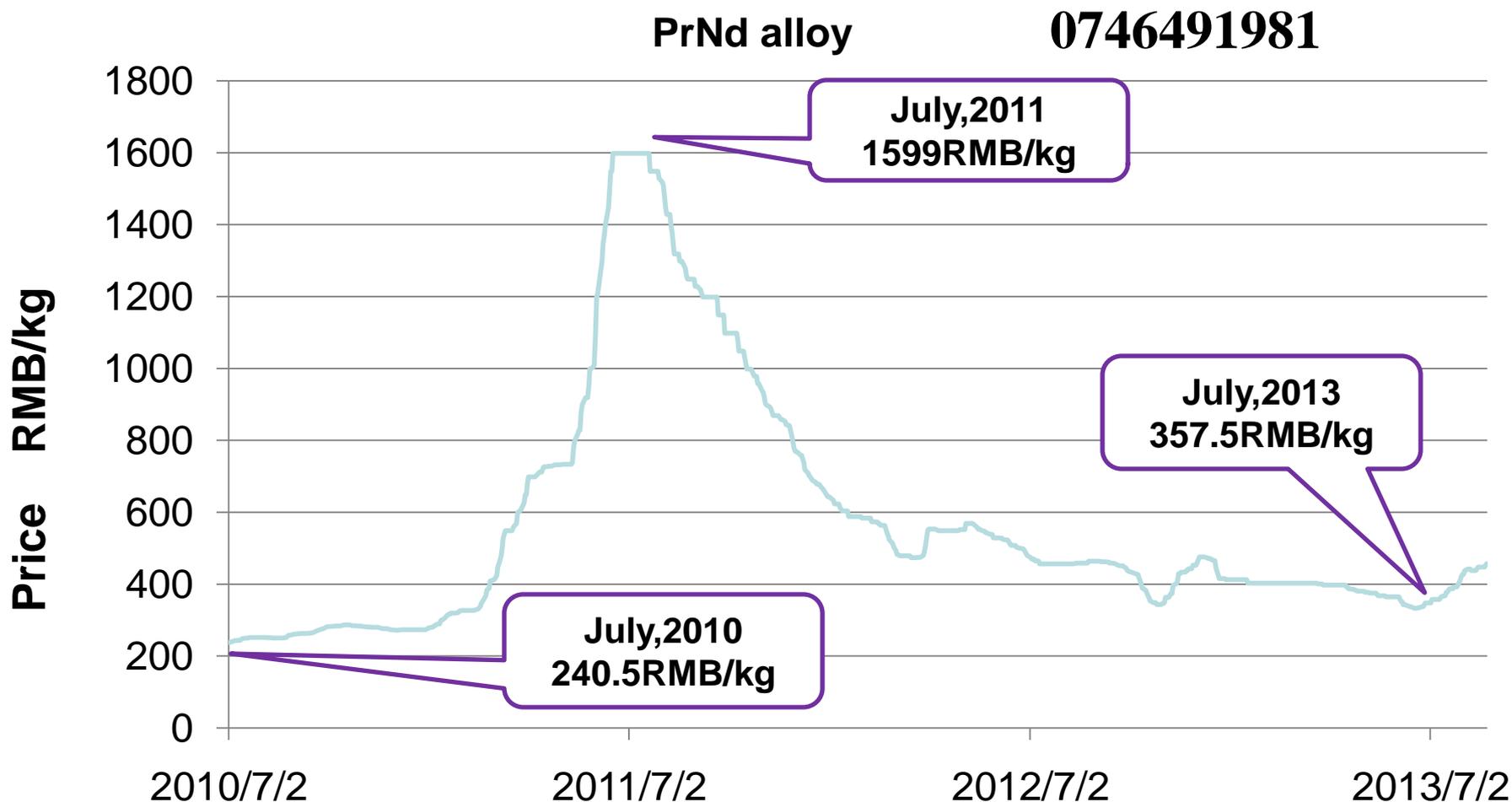


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HPM= rare-earth based magnets
China manages about 96 % of
rare-earth resources in 2011



REE price fluctuations

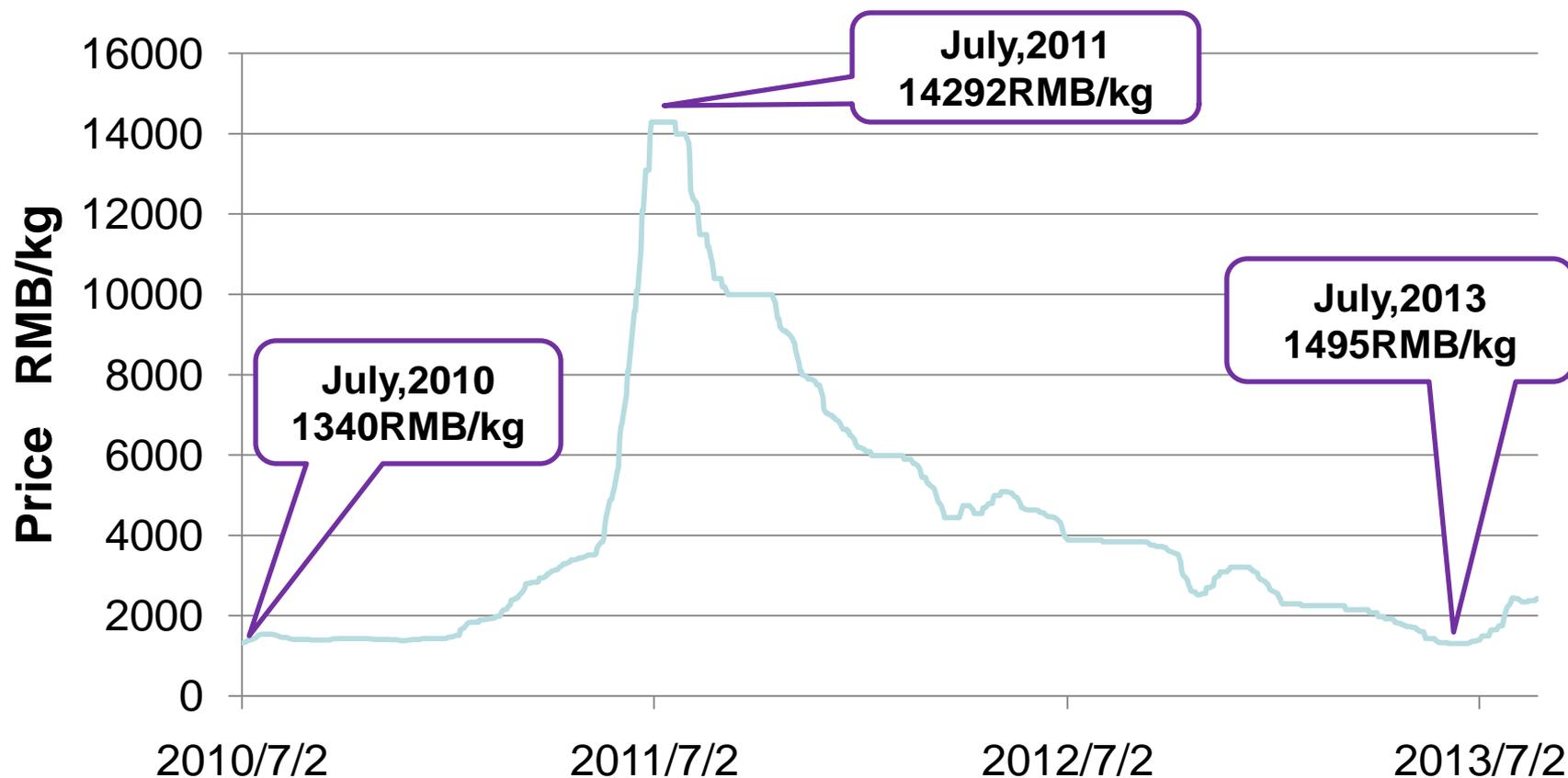


The price of PrNd alloy have increased about 565% form November 2010 to July 2011. In July 2013, it drop to 22.4% compare with top price.

Instability of RE market, ex. Nd: 150 \$/kg/2013; 450 \$/kg/2011, 15 \$/kg/2009

REE price fluctuations

DyFe alloy



The price of DyFe alloy have increased about 967% form July 2010 to July 2011. In July 2013, it drop to 10.46% compare with top price.

Solutions ?

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Yes, we are obliged to have solutions !

1. The increase of usage efficiency.
2. Recycling.

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3. *New magnetic phases* without rare-earth with high magnetic properties for applications as permanent magnets and magnetic refrigeration.: Fe-Co si Fe-Ni tetragonal, Fe-Co ternary or quaternary, Fe_{16}N_2 , MnBi, MnAl, Mn_3Ga , Heusler alloys
4. Soft/hard nanocomposite magnets → *Spring magnets*

Rare Earth-Free Permanent Magnets ?

- ✧ **RE-free** hard magnetic compounds exist: FePt, CoPt, MnBi, MnAl, Zr₂Co₁₁, ε-Fe₂O₃
- ✧ Even the Alnico-type magnets still have a room for improvement; their theoretical $(BH)_{\max}$ is 36-49 MGOe and they have excellent temperature stability; **Artificial Alnicos!**

Compound	Structure	Saturation magnetization	Curie temperature (°C)	Anisotropy constant K_1 (MJ/m ³)	$(BH)_m$ (MGOe)
Co	hexagonal	17.6 kG	1115	0.53	
FePt	tetragonal	14.3 kG	477	6.6	
CoPt	tetragonal	10.0 kG	567	4.9	
Co ₃ Pt	hexagonal	13.8 kG	727	2.0	
MnAl	tetragonal	6.2 kG	377	1.7	9.6
MnBi	hexagonal	7.8 kG	357	1.2	16-17
BaFe ₁₂ O ₁₉	hexagonal	4.8 kG	450	0.33	3-4
Zr ₂ Co ₁₁	orthorhombic(?)	≈70 emu/g	500	? ($H_A = 34$ kOe)	14
ε-Fe ₂ O ₃	orthorhombic	≈16 emu/g	?	? ($H_c = 23.4$ kOe)	
Alnico	Cubic (shape)	12-14			8-11(36)
SmCo ₅	hexagonal	11.4 kG	681	17.0	25-30
Nd ₂ Fe ₁₄ B	tetragonal	16.0 kG	312	5.0	30-57

Solutions ?

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Our recent research in these directions

1. *New magnetic phases* without rare-earth with high magnetic properties
2. **Soft/hard** nanocomposite magnets → *Spring magnets*

Our recent research in these directions

1. New magnetic phases without rare-earth with high magnetic properties

✓ *MnBi*

✓ *MnAl*

• $\text{Mn}_{50+\delta}\text{Al}_{50-\delta}$; $\delta=4$

• $\text{Mn}_{50}\text{Al}_{50-\delta}\text{X}_\delta$; $\delta=4$, X=Ni, Zn, Ti

2. **Soft/hard** nanocomposite magnets → Spring magnets

✓ hard magnetic phases of **SmCo_5 , SmCo_3Cu_2 , $\text{R}_2\text{Fe}_{14}\text{B}$**

✓ soft magnetic phases of **$\alpha\text{-Fe}$, Fe-Co (~20 or 10 wt%)**

Our recent research in these directions

1. New magnetic phases without rare-earth with high magnetic properties

✓ *MnBi*

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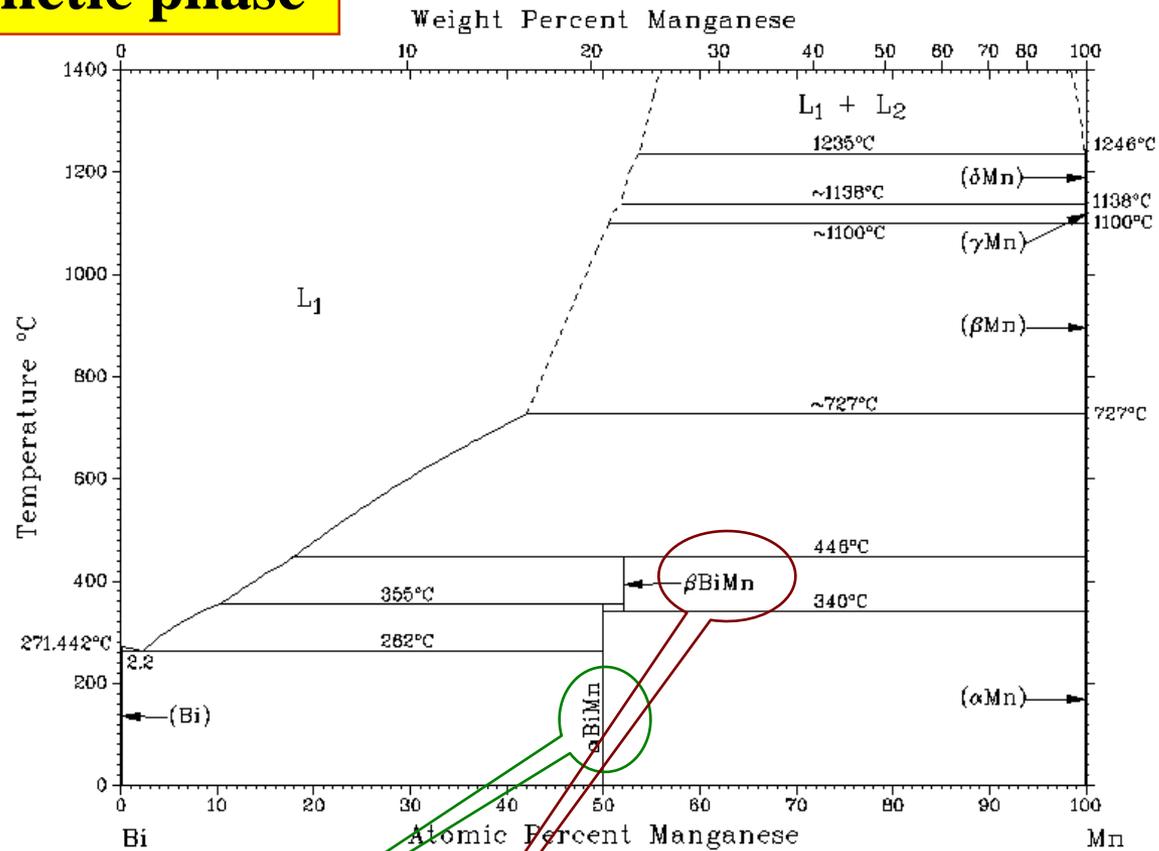
2. **Soft/hard** nanocomposite magnets → Spring magnets

- ✓ hard magnetic phases of **SmCo₅, SmCo₃Cu₂, R₂Fe₁₄B**
 - **SmCo₅** large anisotropy
 - **SmCo₃Cu₂** large coercivity
 - **R₂Fe₁₄B** best magnets
- ✓ soft magnetic phases of **α-Fe, Fe-Co (~20 or 10 wt%)**

Outline

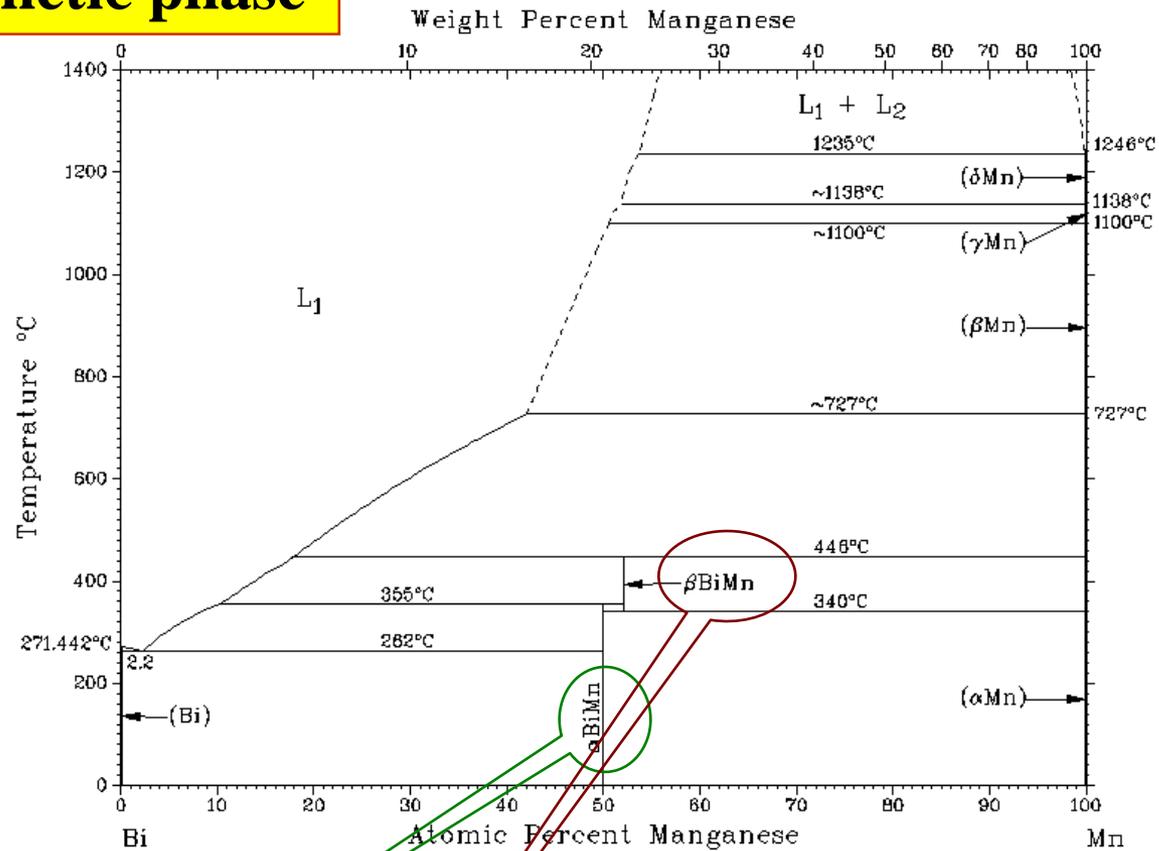
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MnBi magnetic phase



- Low temperature phase (LTP)-NiAs type Hexagonal (Ferromagnetic)
- High temperature phase (HTP) -Distorted Ni_2In type hexagonal (paramagnetic)

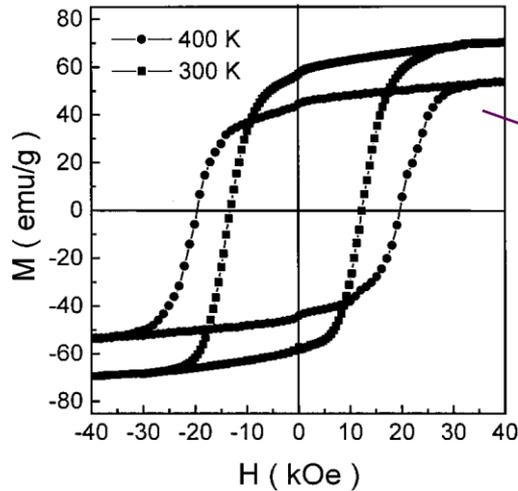
MnBi magnetic phase



- Low temperature phase (LTP)-NiAs type Hexagonal (Ferromagnetic)
- High temperature phase (HTP) -Distorted Ni_2In type hexagonal (paramagnetic)
- Quenched high temperature phase (QHTP)-Orthorhombic (Ferromagnetic-low M_s)

Some previous works in MnBi magnetic compound*

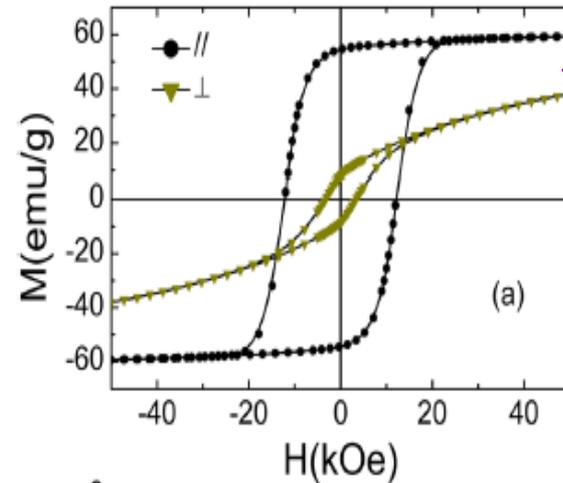
Powders: Sintered method & magnetic separation



$(BH)_{\max} = 7.7 \text{ MGOe}$
(powders)

J B Yang et al. *J. Phys.: Condens. Matter* 14 (2002) 6509

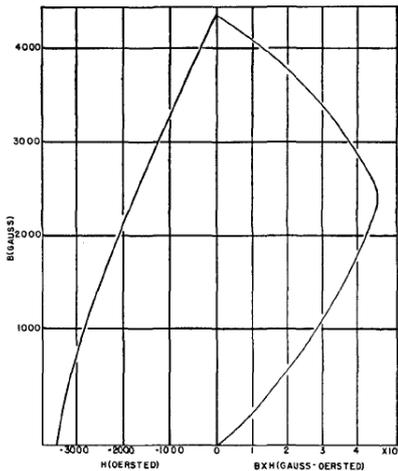
Nanocrystalline MnBi by melt-spinning technique



$(BH)_{\max} = 7.1 \text{ MGOe}$
(powders of melt-spun ribbons)

Yang et al. *Appl. Phys. Lett.* 99, 082505 (2011)

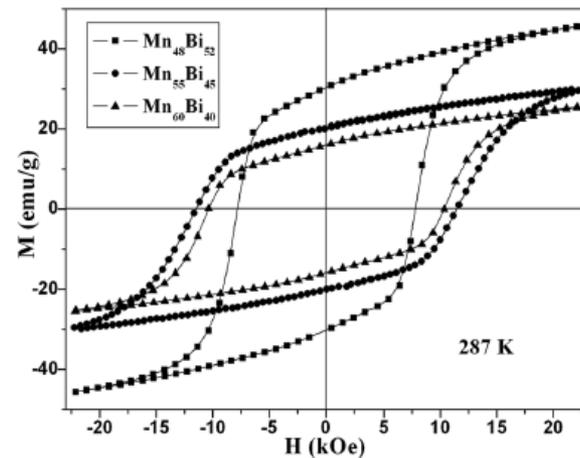
Hot pressed bulk magnet



$(BH)_{\max} = 4.3 \text{ MGOe}$
Density: 90%

Adams et al. *J. Appl. Phys.* 23 1207 (1952)

Spark plasma sintered bulk magnet

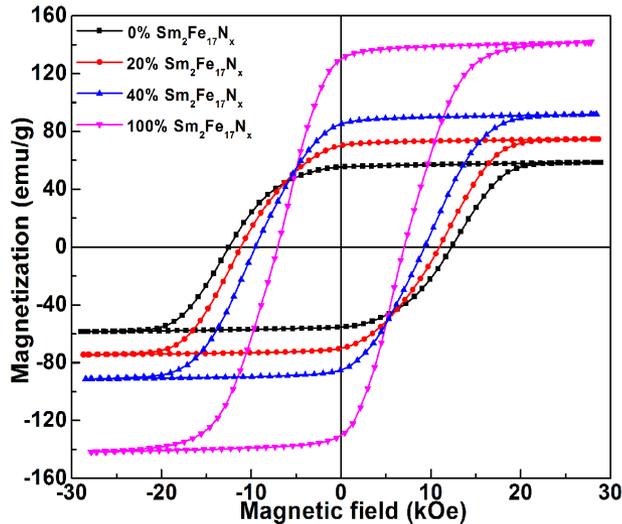


$(BH)_{\max} < 2 \text{ MGOe}$
Density: 93%

Zhang et al. *J. Appl. Phys.* 109, 07A722 (2011)

*G. Hadjipanayis, Delaware University, Energy and Materials Criticality Workshop, Santorini 2013

Magnetic properties of MnBi/Sm₂Fe₁₇N_x nanocomposite powders



- ❖ MnBi powders: H_c of 12.4 kOe with M_r of 55 emu/g
- ❖ Sm₂Fe₁₇N_x powders: H_c of 7 kOe with M_r of 130
- ❖ Hybrid magnet powders exhibit
 - H_c and M_r values intermediate to those of pure MnBi and Sm₂Fe₁₇N_x
 - *anisotropic magnetic characteristics with M_r/M_s ratio greater than 0.91*

Our main results

MnBi

Synthesis:

- ✓ [melting](#): Mn and Bi of 99,99% purity (1 wt % Mn in excess)
- ✓ [annealing](#): 258-420°C/ from 2 hours to 4 days
- ✓ [mechanical milling](#): of bulk MnBi phase for 2 hours

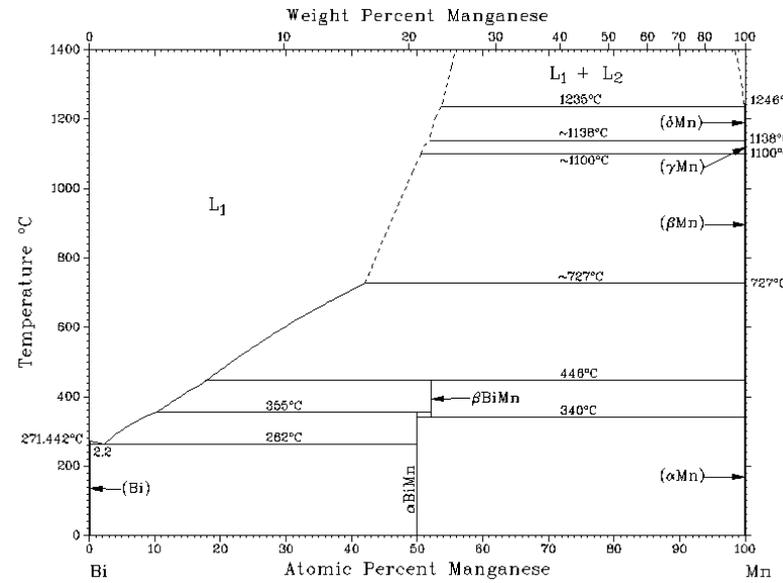
X-rays powder diffraction:

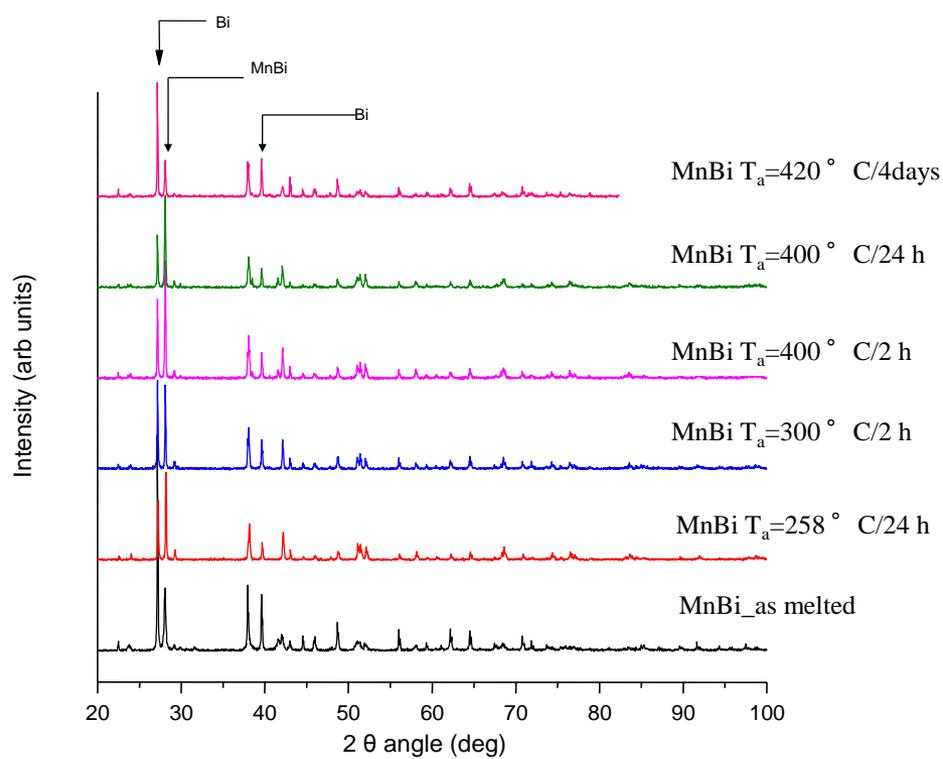
- ✓ K_{α} radiation of copper in the angular range $2\theta = 20 - 100^{\circ}$ and
- ✓ $K_{\alpha 1}$ radiation of cobalt in angular range $2\theta = 20 - 80^{\circ}$

Magnetic measurements:

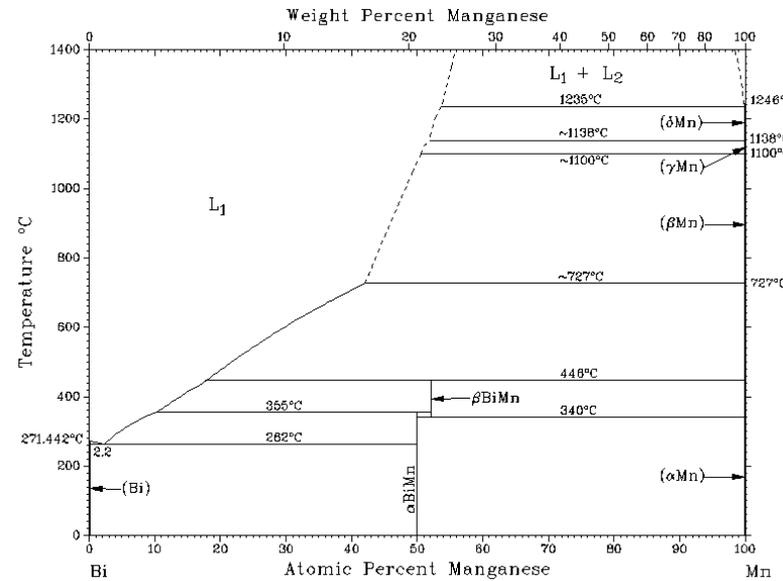
- ✓ [extraction method](#) in a continuous magnetic field of up to ± 10 T

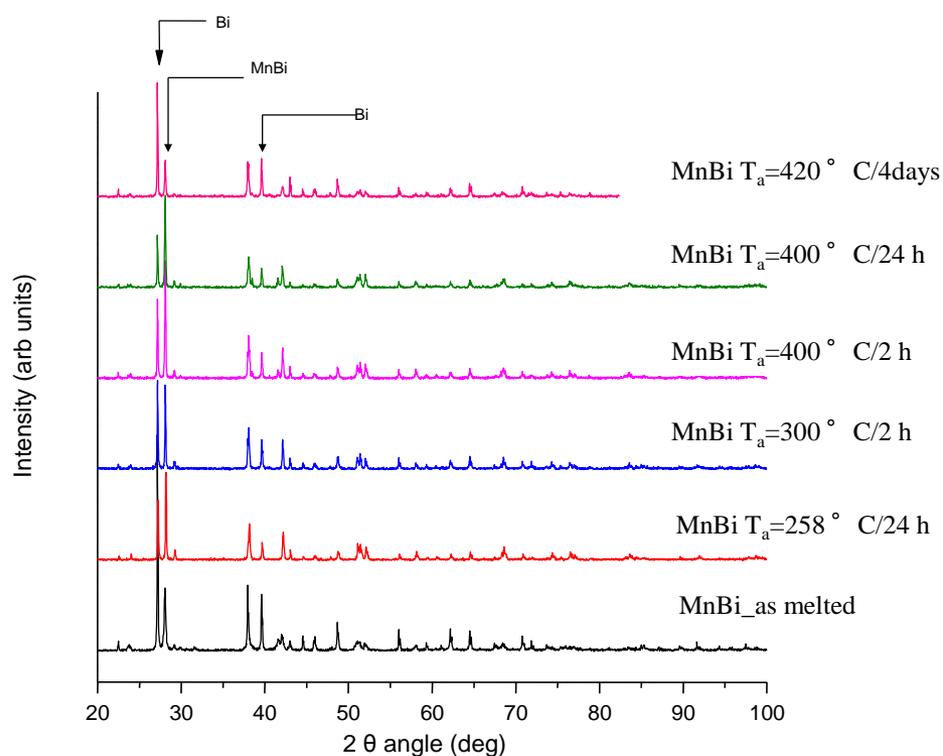
MnBi: influence of *annealing*



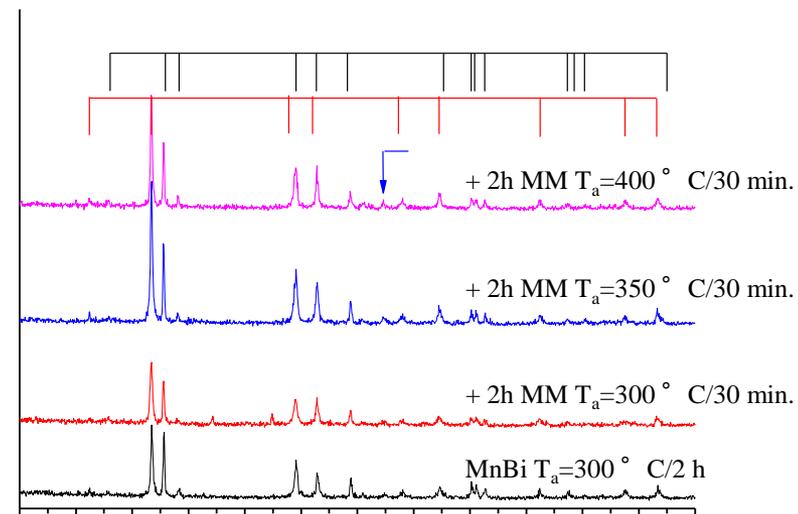


MnBi: influence of *annealing*; XRD Cu K_α radiation

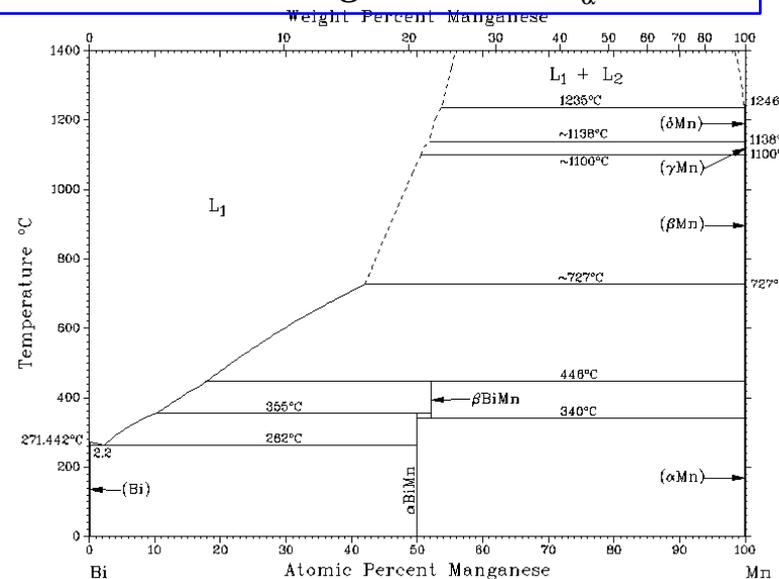


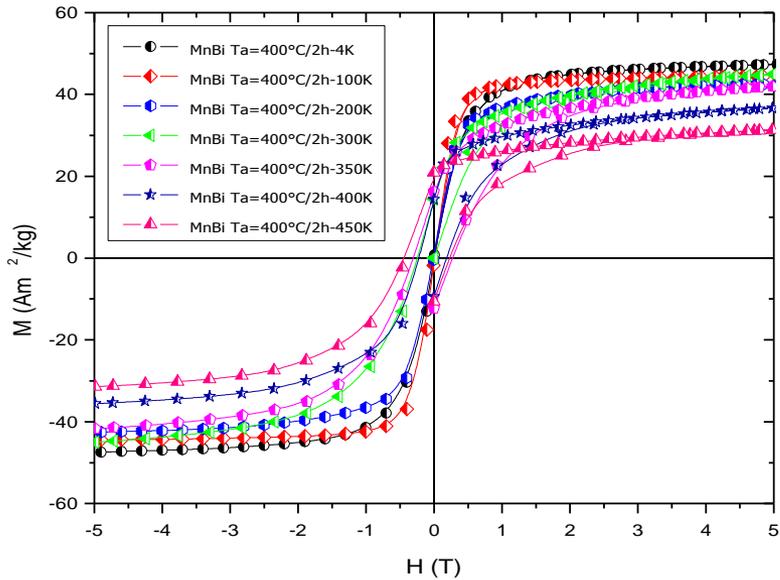


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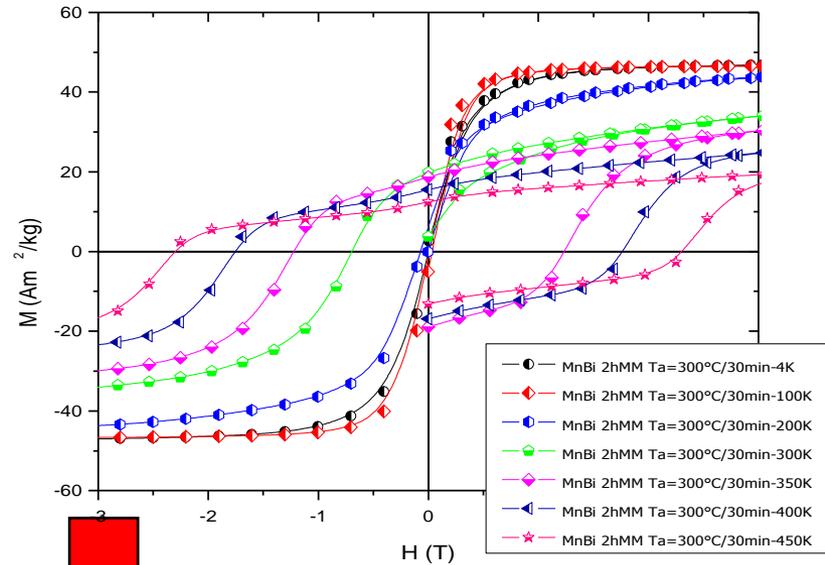


MnBi: influence of *milling*; XRD Co K_α radiation





M(H) for different T,
MnBi melted samples, annealed at 400°C for 2 h



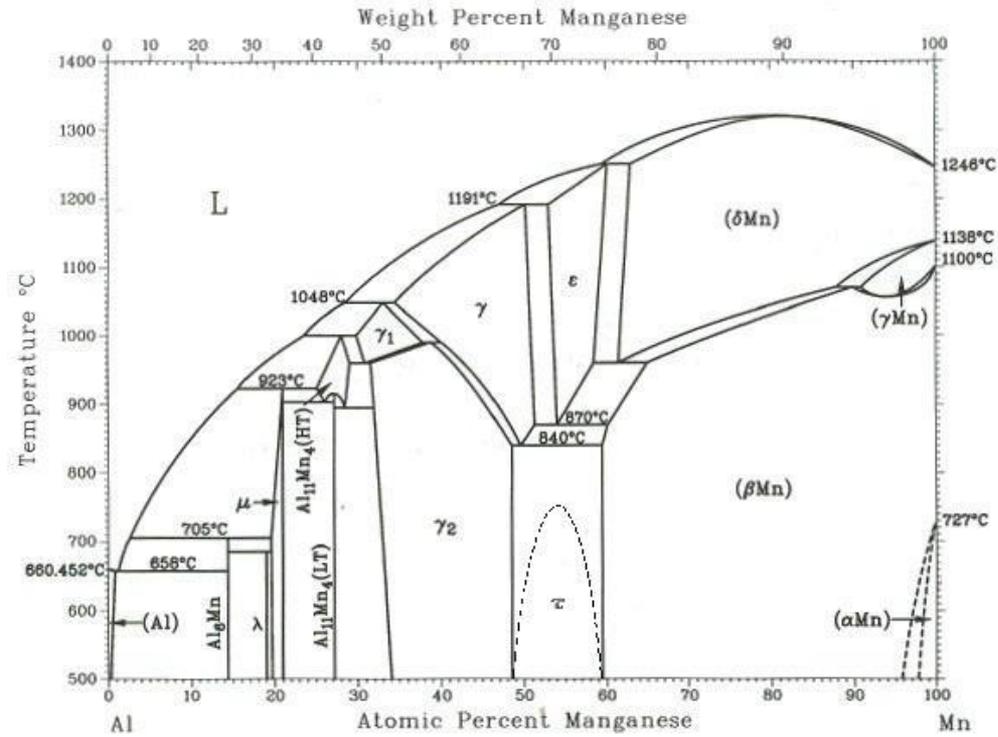
Hysteresis loops for different T,
MnBi 2h MM+TT 300 °C/30 min

**Important coercivity at high temperature,
competition with REPM**

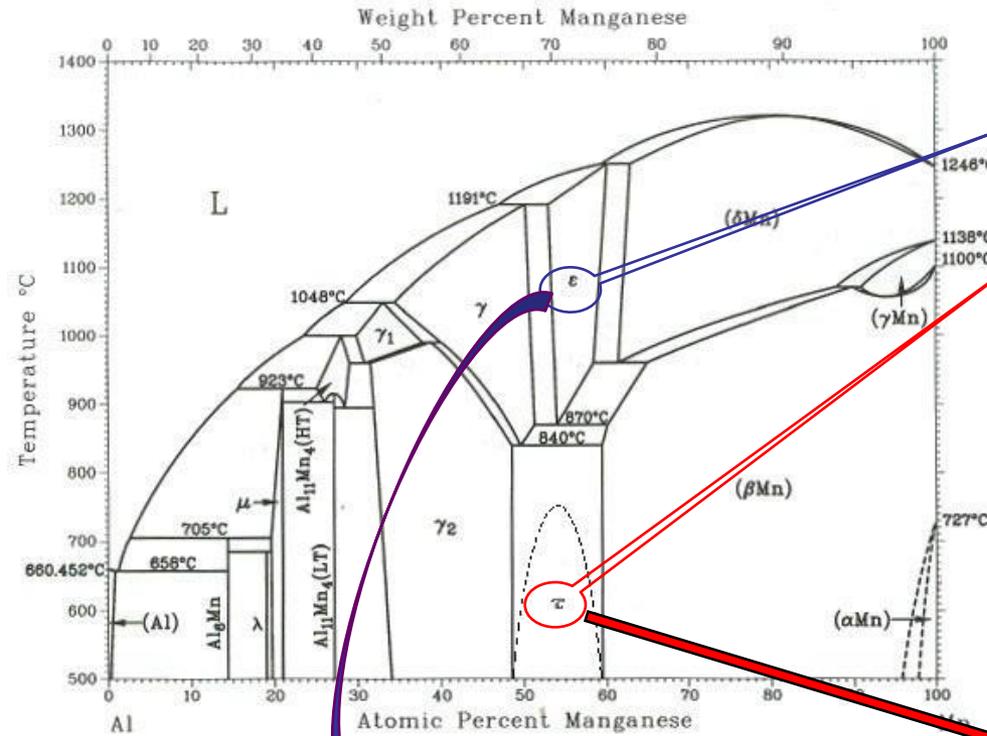
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Mn-Al magnetic phase

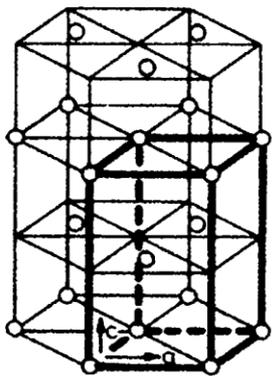


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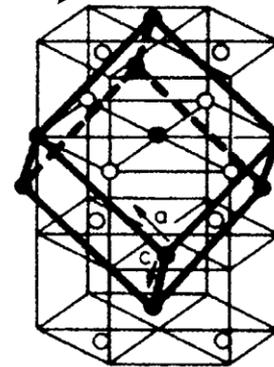


ε phase –
antiferromagnetic hexagonal structure

τ phase –
L1₀-ferromagnetic tetragonal structure

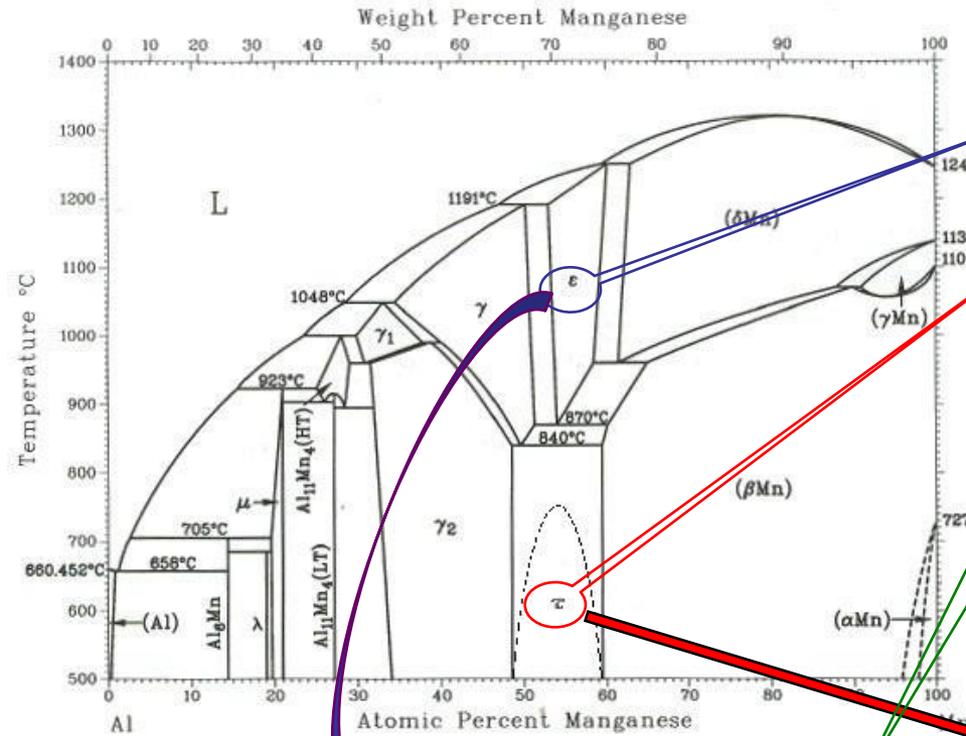


A₃ (ε)



L1₀ (τ)

Mn-Al magnetic phase

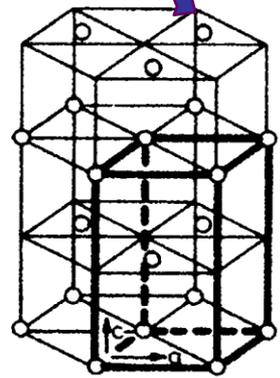


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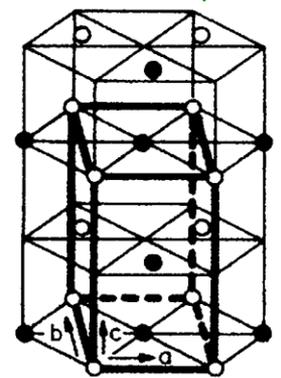
ε' phase –
Intermediate ferromagnetic ordered orthorhombic phase

Mn₅₄Al₄₆ - the best results



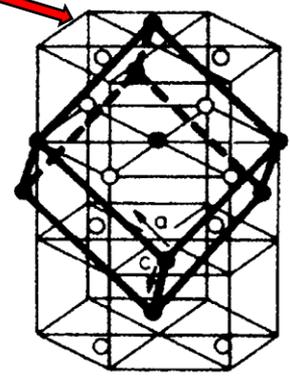
A3 (ε)

ordering
reaction



B19 (ε')

martensitic
shear



L1₀(τ)

- Phase transformation proposed by Broek et. al. with intermediate ordered orthorhombic phase (B19) denoted by ε'.
- Acta Metall.,27(1979) 1497

Phase formation, microstructure, magnetic properties of the Mn–Al–C; bulk or MM*

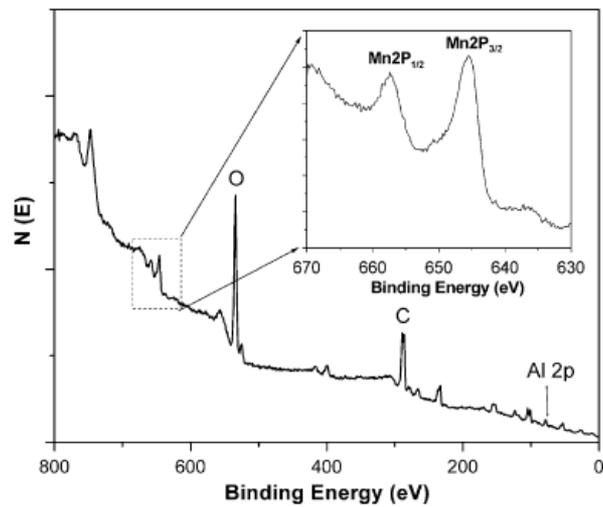


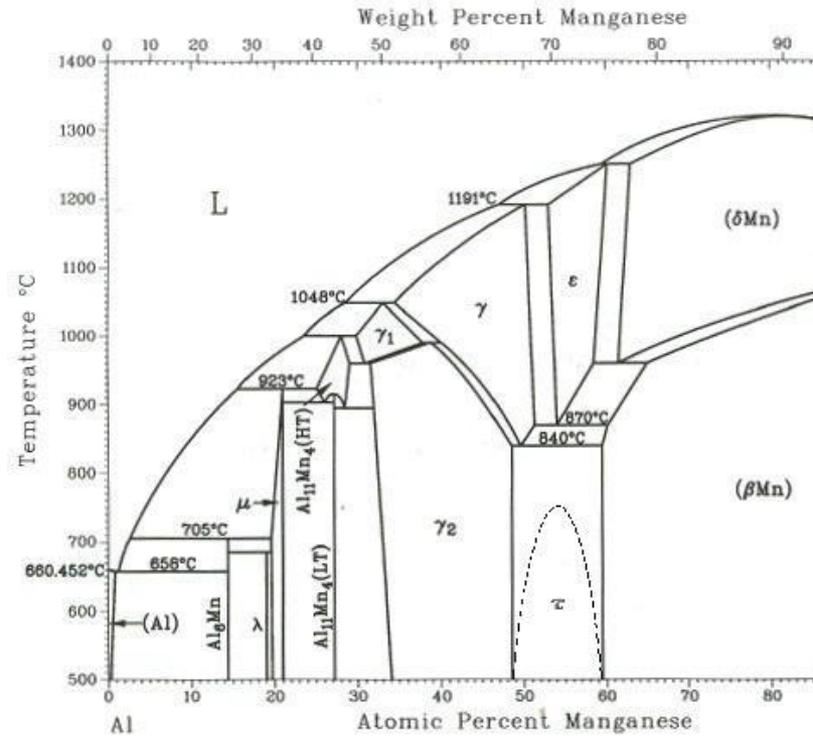
Fig. 4. XPS spectrum of an annealed Mn₅₄Al₄₆. The inset is the corresponding spectrum of Mn2p.



τ -phase nucleates almost exclusively at the ϵ -phase grain boundaries

*Q. Zeng, I. Baker, J.B. Cui, Z.C. Yan, JMMM, 308 (2007) 214–226

Phase formation, microstructure, magnetic properties of the Mn–Al–C; bulk or MM*



- DSC: the transformation $\varepsilon(\varepsilon') \rightarrow \tau$ phase,
- τ -phase stabilized by C doping,
- C doping *cannot prevent* the formation of the equilibrium phases from the metastable ε -phase during annealing.

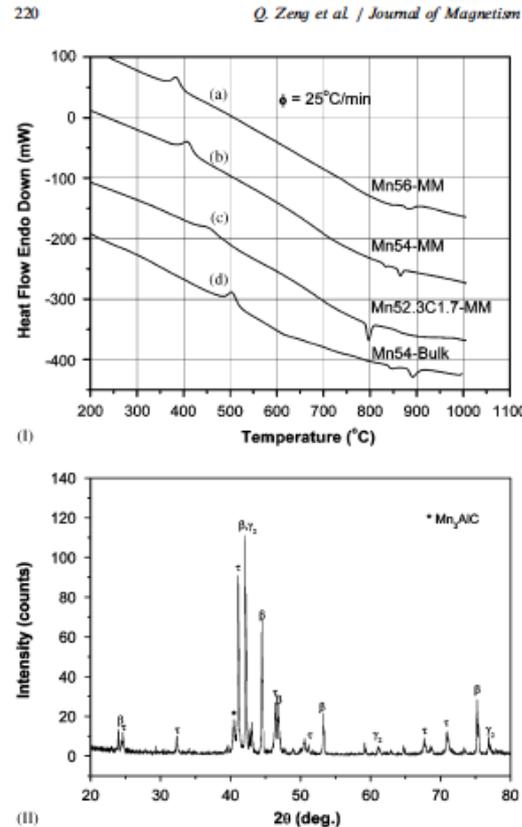


Fig. 7. (I) DTA curves for (a) MM Mn₅₆Al₄₆, (b) MM Mn₅₄Al₄₆, (c) Mn_{52.3}Al₄₆C_{1.7}, and (d) bulk Mn₅₄Al₄₆. The heating rate was 25°C/min. (II) XRD pattern for MM Mn_{52.3}Al₄₆C_{1.7} heated to 760°C at the heating rate of 25°C/min and quenched.

Q. Zeng et al. / Journal of Magnetism and Magnetic Materials 308 (2007) 214–226

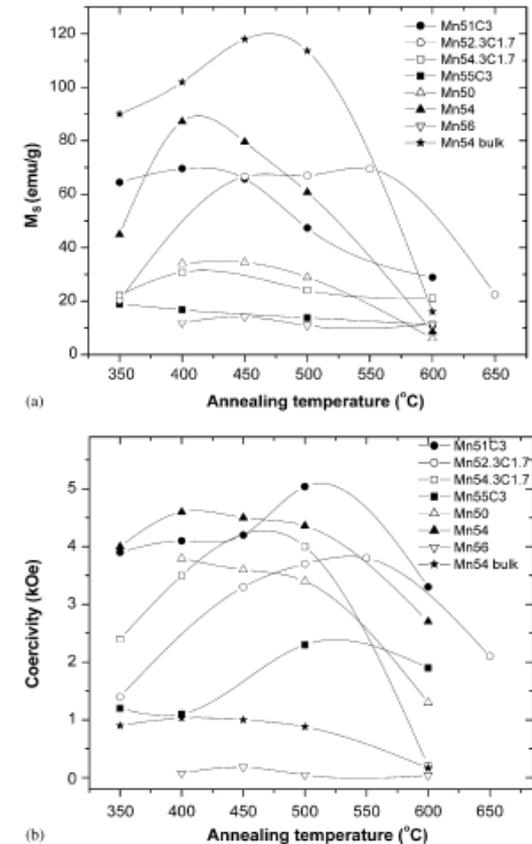


Fig. 8. Dependence of M_S and H_C on the annealing temperatures for MM and bulk samples with various compositions as indicated. The annealing time was 30 min.

Phase formation, microstructure, magnetic properties of the Mn–Al–C; bulk or MM*

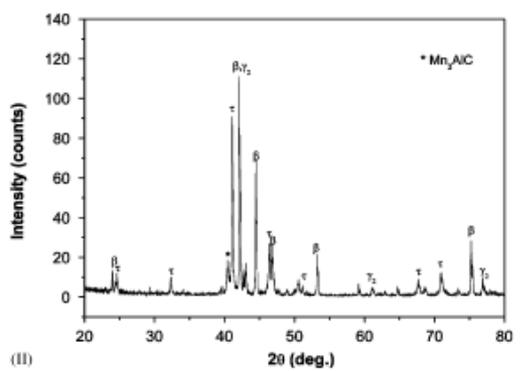
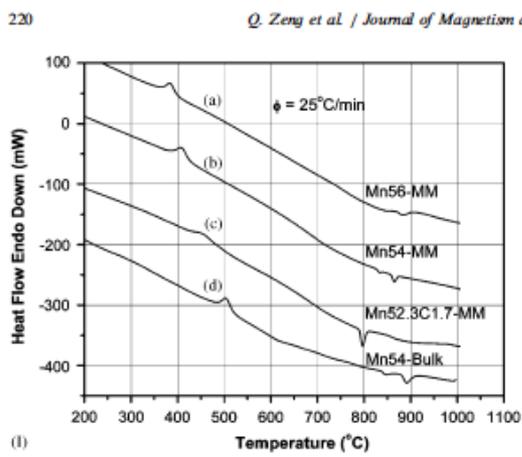
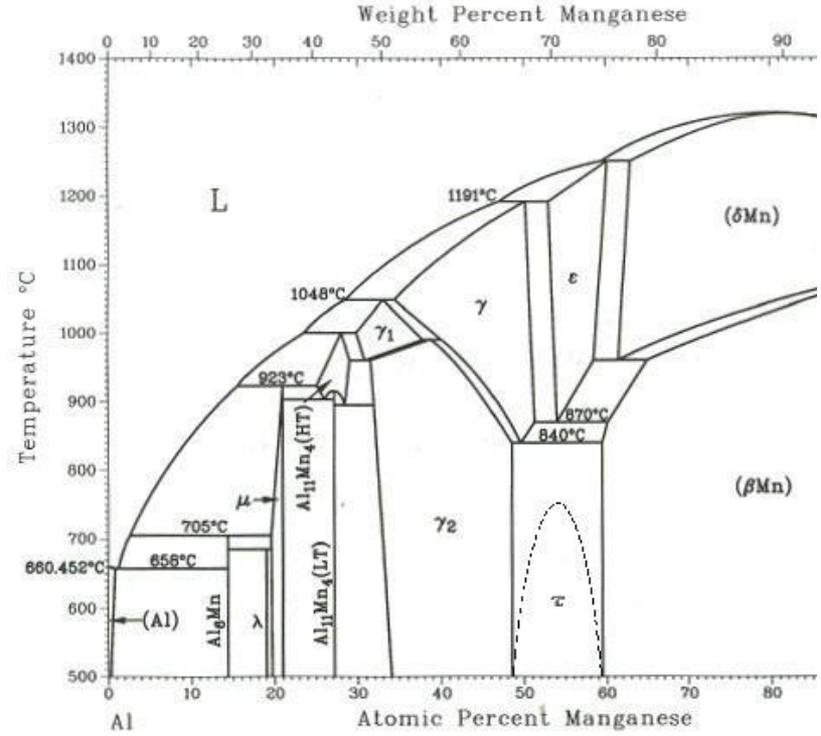


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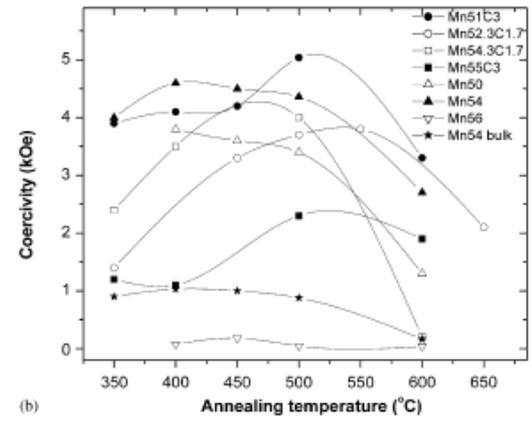
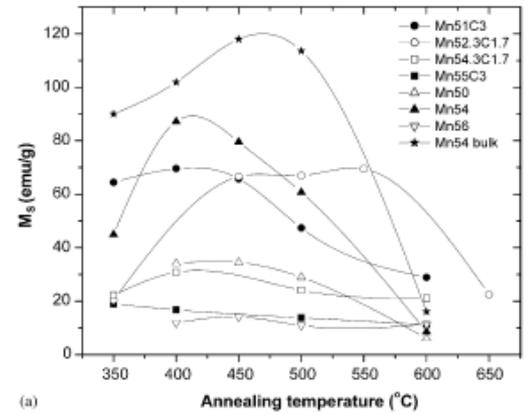
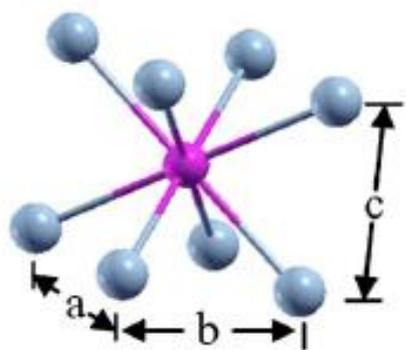


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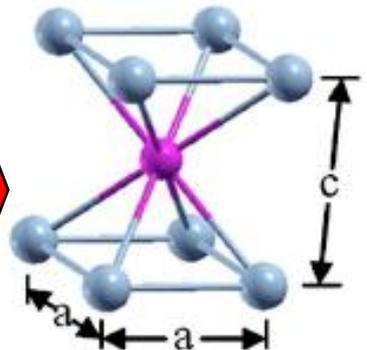
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The optimal magnetic properties for the MM samples, $H_c=4.8$ kOe, $M_r=45$ emu/g and $M_s= 89$ emu/g, were obtained for Mn₅₄Al₄₆ annealed at 400 ° C for 10 min

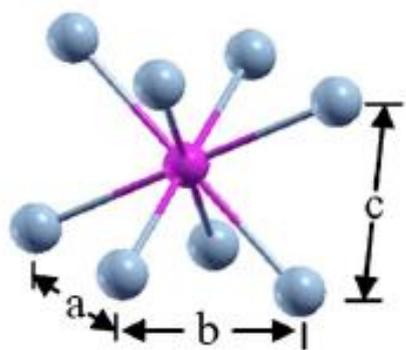
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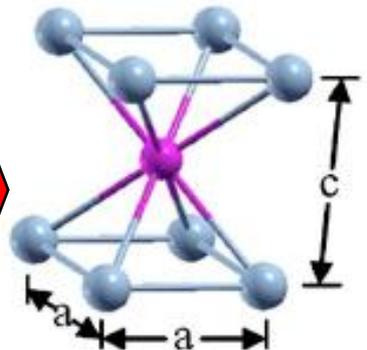
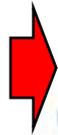
ϵ' -phase



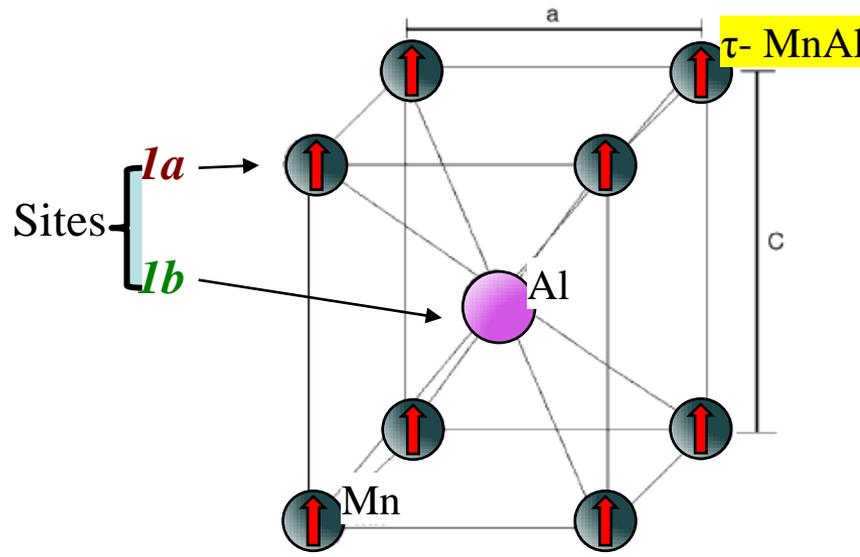
τ -phase.



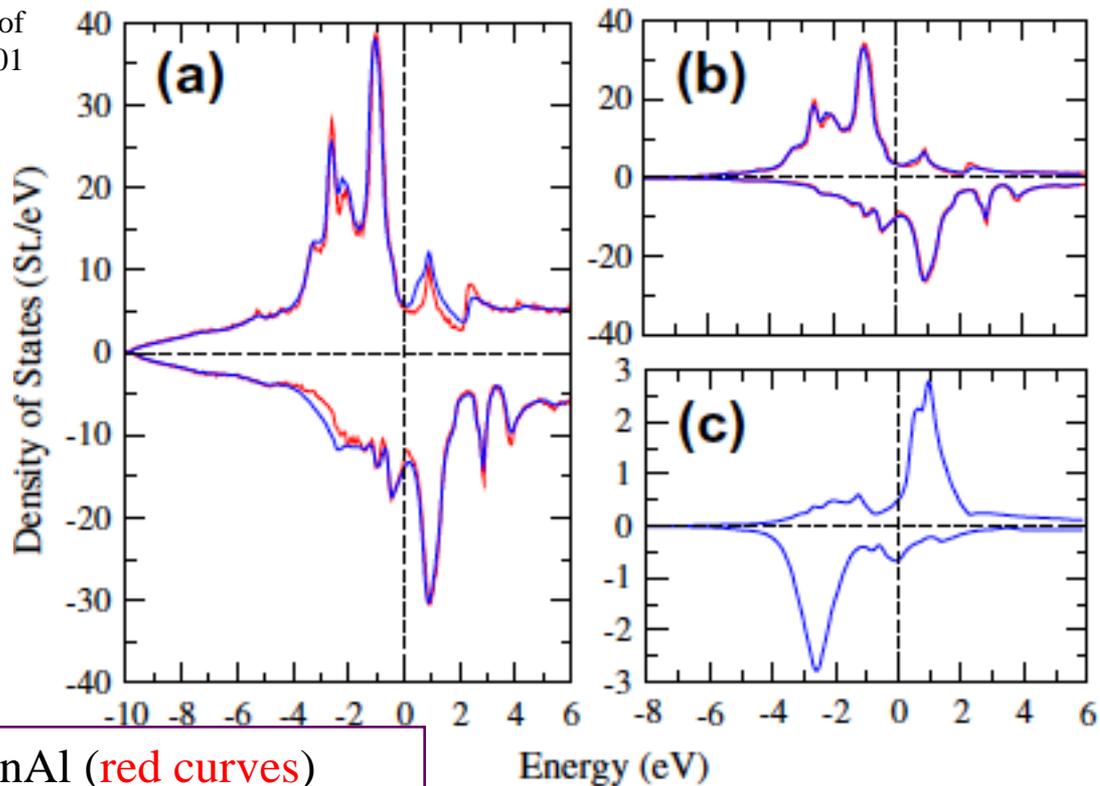
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τ -phase.



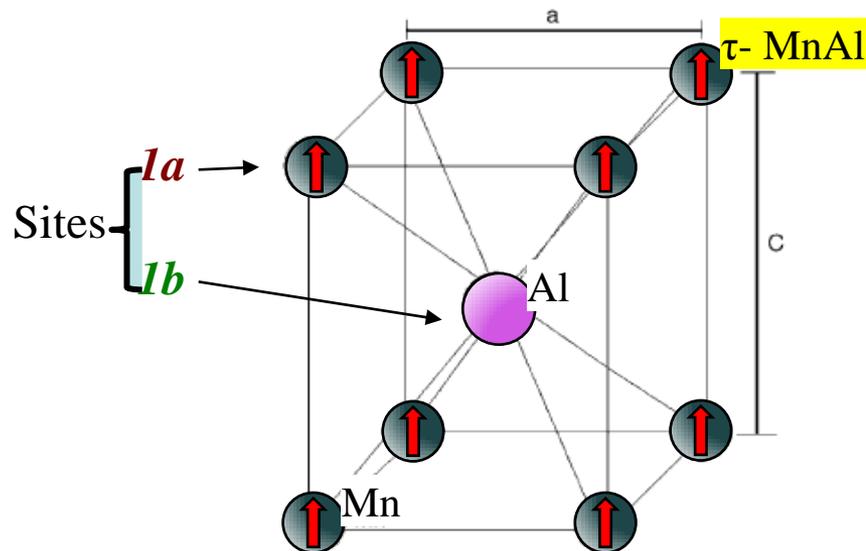
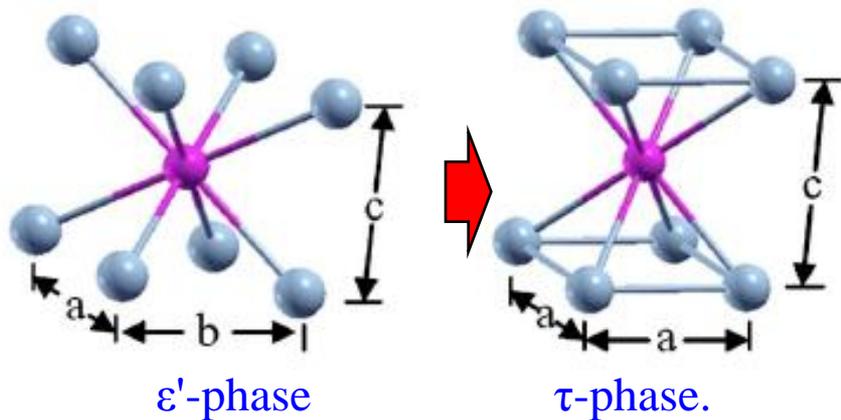
(a) the total density of states



(b) the 3d partial density of states of Mn at the *1a* site

(c) the 3d partial density of states of Mn at the *1b* site of the P4/mmm unit cell

stoichiometric MnAl (red curves)
 $\text{Mn}_{50+\delta}\text{Al}_{50-\delta}$; $\delta=4$ (blue curves).



Our results

Mn₅₀Al₄₆Ni₄ and Mn₅₄Al₄₆

- Ingots were prepared by arc or induction melting
- Different heat treatment to stabilize the desired magnetic phase

Measurements

- Differential thermal analysis (DTA)
- XRD on Brüker D8 Advance diffractometer
- Thermomagnetic measurements up to 800 K

DTA $\text{Mn}_{50}\text{Al}_{46}\text{Ni}_4$ and $\text{Mn}_{54}\text{Al}_{46}$

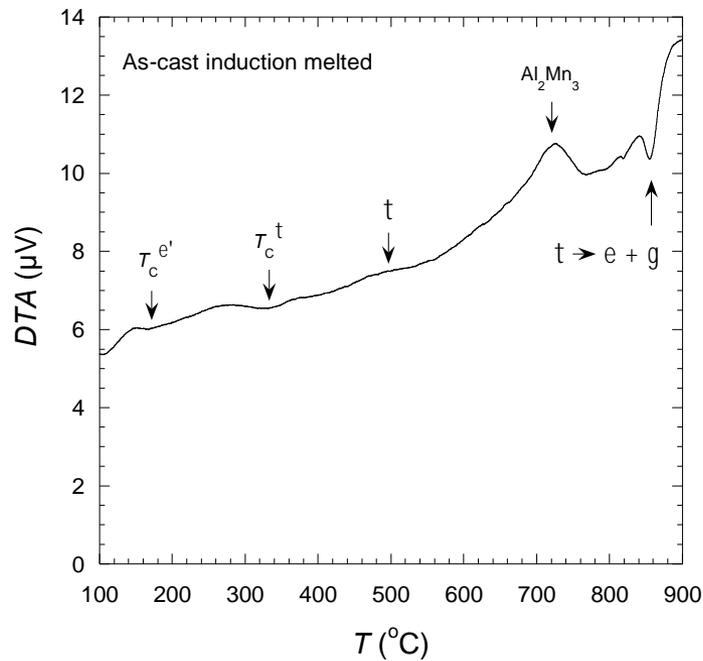
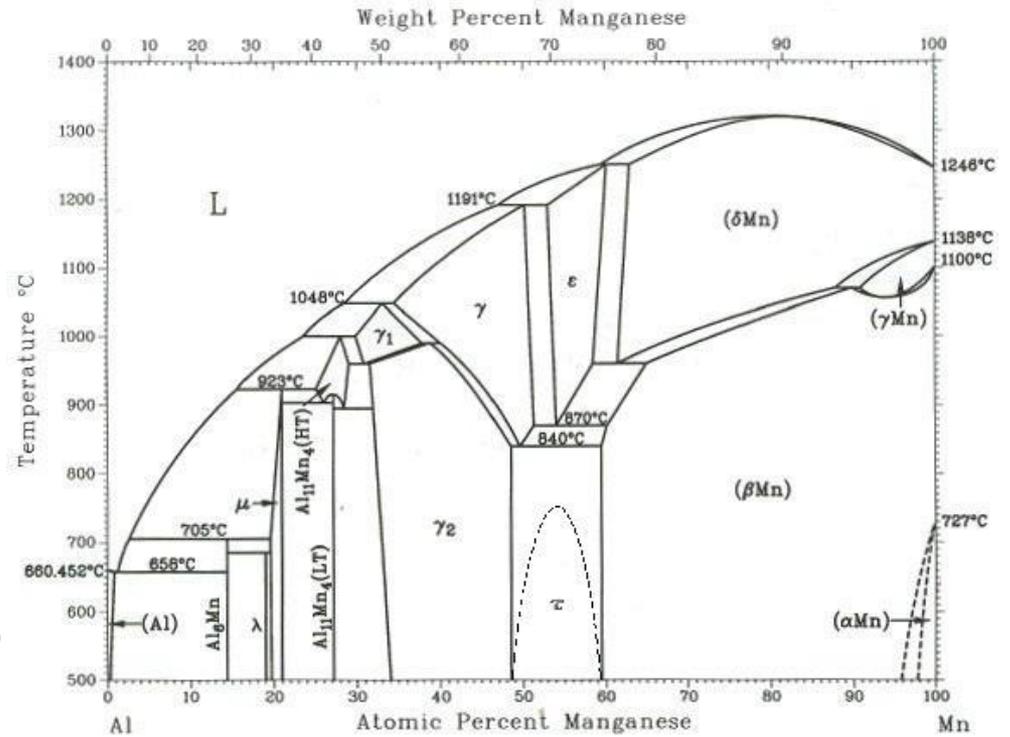
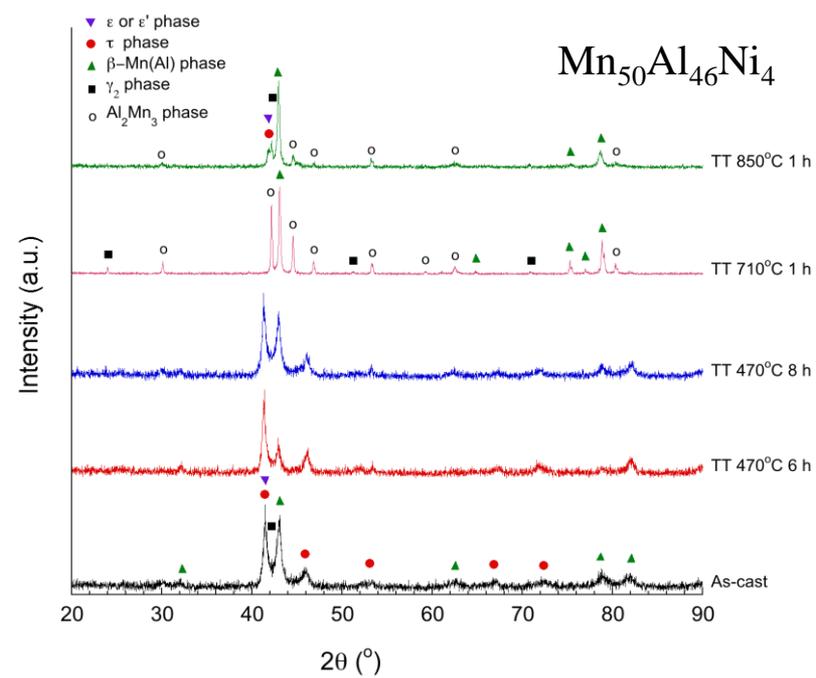
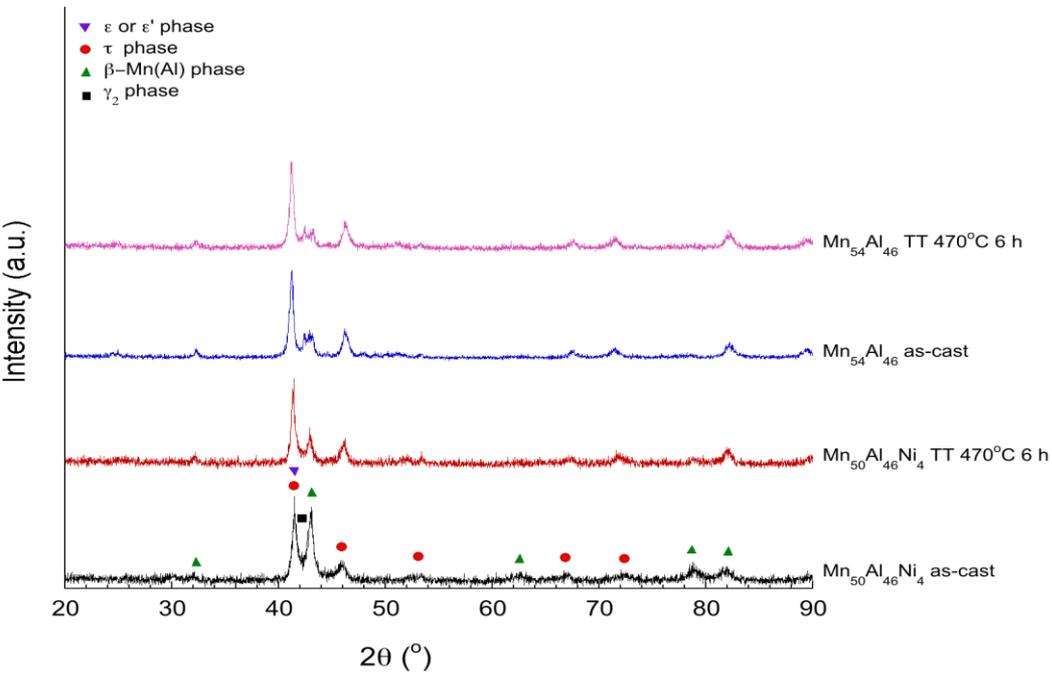


Fig. 1. DTA curve for the as-cast $\text{Mn}_{50}\text{Al}_{46}\text{Ni}_4$ alloy.

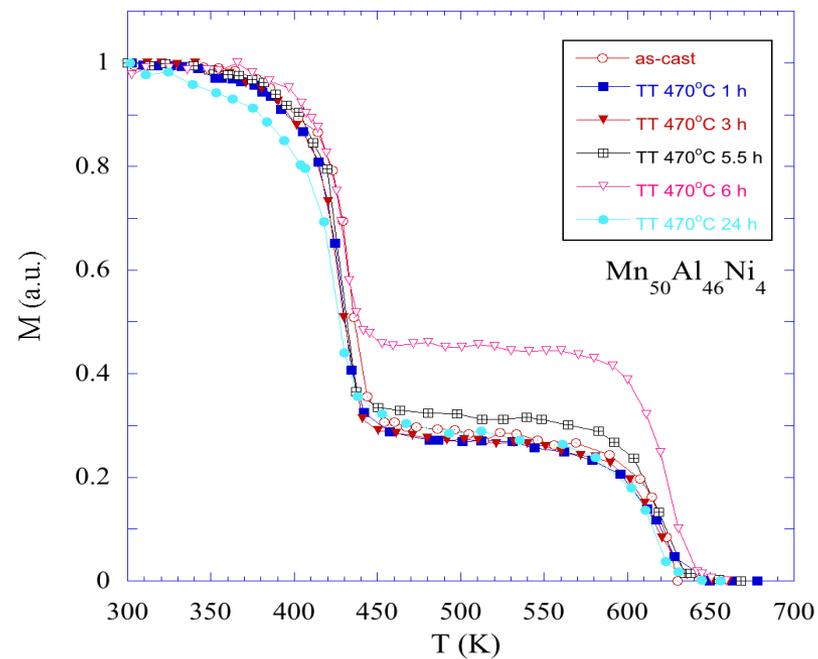
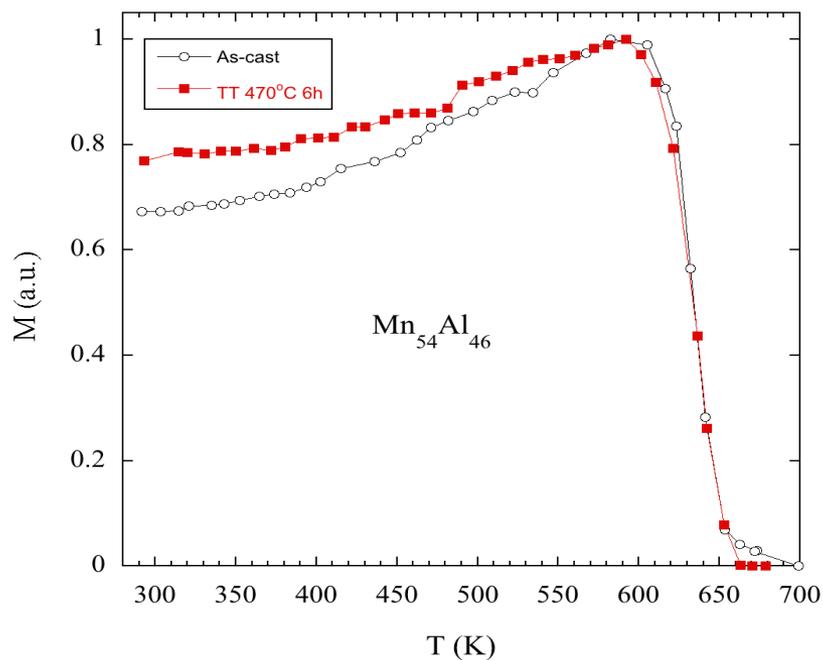
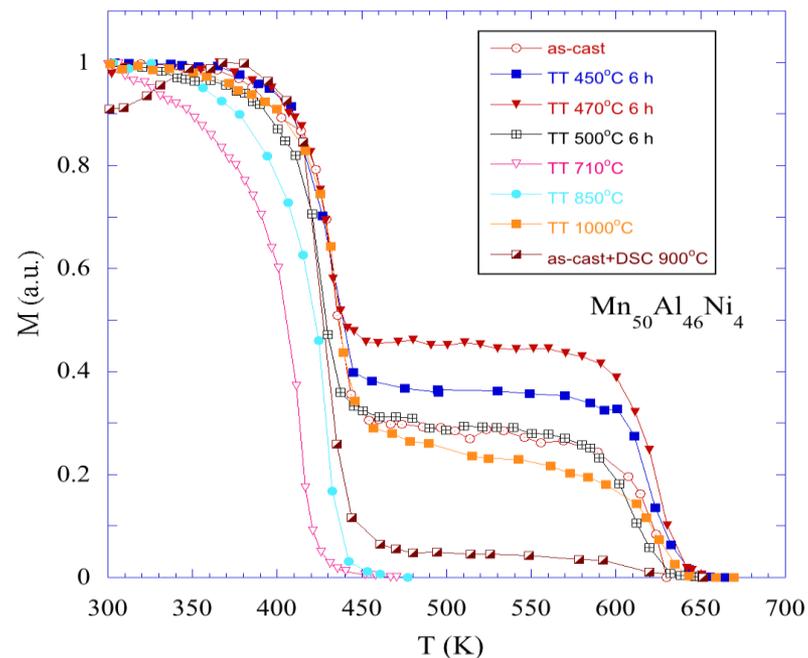


XRD $\text{Mn}_{50}\text{Al}_{46}\text{Ni}_4$ and $\text{Mn}_{54}\text{Al}_{46}$

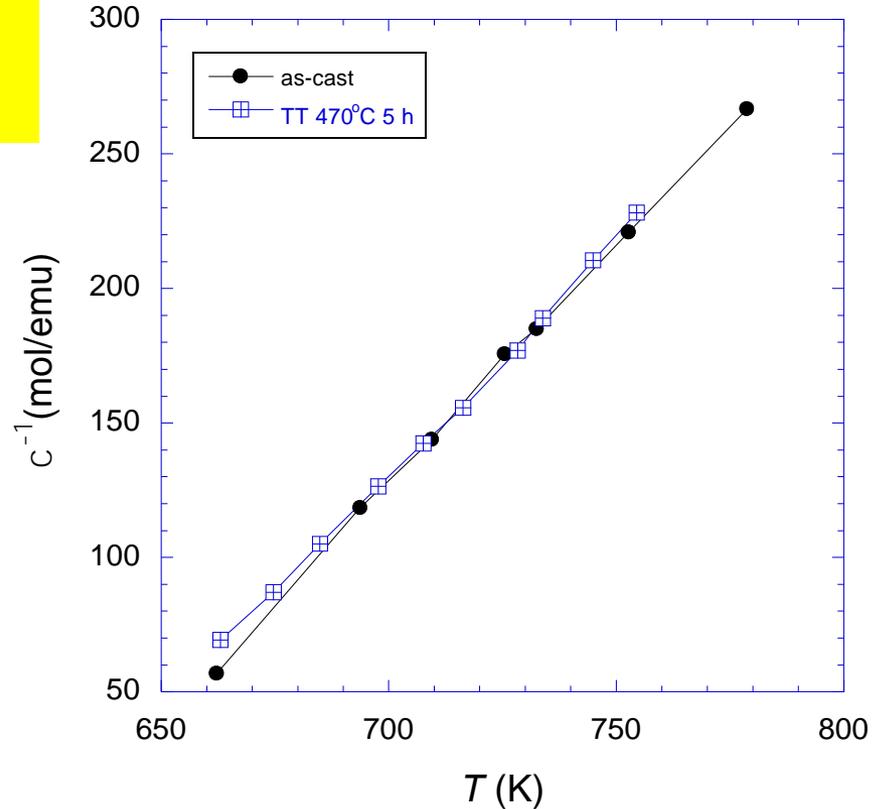


Thermomagnetic studies

$\text{Mn}_{50}\text{Al}_{46}\text{Ni}_4$ and $\text{Mn}_{54}\text{Al}_{46}$



Paramagnetic behavior



The effective magnetic moments μ_{eff} , the spins S , of the Mn atoms, the τ -phase content c_{τ} , the average Mn moments in the ordered state μ_{exp} , the paramagnetic Curie temperatures θ , and the Curie temperatures T_C , of the τ -phase of the $\text{Mn}_{50}\text{Al}_{46}\text{Ni}_4$ alloy.

Sample	$\mu_{\text{eff}}/\text{Mn}$ (μ_B)	S_{Mn}	c_{τ} (%)	μ_{exp} (μ_B)	θ (K)	T_C (K)
As-cast $\text{Mn}_{50}\text{Al}_{46}\text{Ni}_4$	2.98	1.07	35	2.06	629	624
$\text{Mn}_{50}\text{Al}_{46}\text{Ni}_4$ annealed at 470 °C for 5h	3.04	1.1	43	2.09	624	635

Abstract: In this study we present the results of electronic structure calculations and the experimental investigations on the structural and magnetic properties of the $Mn_{50}Al_{46}M_4$ ($M = Mn, Ni, Ti$) alloys. The highest magnetic moment was found for the $Mn_{50}Al_{46}Ni_4$ alloy. Total energy calculations point to a more stable antiferromagnetic configuration of the $Mn^{1+}Mn^{2+}$ pair. DTA measurements pointed out the formation of the metastable τ phase around 470 °C and its decomposition into the stable τ_2 and β -Mn(Al) phases at 850 °C. The τ phase was found along with the ϵ' phase only in the as-cast sample and the ones annealed at 470 °C. A maximum τ phase concentration of 50% was found for the sample annealed at 470 °C for 6 h. The close values of the theoretical and experimental effective magnetic moments confirm the existence of the ϵ' and τ phases in these alloys.

Experimental and Computational Details:

- Electronic structure calculations were performed in the framework of the Local Density Approximation (LDA) of the Density Functional Theory by means of the SPR-KKR method using the experimentally determined lattice parameters for $Mn_{50}Al_{46}$ ($a = 3.94 \text{ \AA}$, $c = 3.58 \text{ \AA}$). All relativistic effects have been taken into account, including the spin-orbit coupling, in the ferromagnetic (FM) and antiferromagnetic (AFM) spin configurations of the $Mn^{1+}Mn^{2+}$ pair.
- The $Mn_{50}Al_{46}$ and $Mn_{50}Al_{46}Ni_4$ ingots were prepared by induction melting of the starting components under a purified Ar atmosphere.
- The samples were annealed in an inert Ar atmosphere at temperatures between 470 and 850 °C for different times followed by quenching in water.
- XRD investigations were performed using a Bruker D8 Advance X-ray diffractometer with $Cu K_{\alpha}$ radiation.
- Differential thermal analysis (DTA) was performed between 100 and 900 °C under Ar atmosphere with a temperature ramp rate of 20 °C/min.
- The magnetization and magnetic susceptibility were measured with a Weiss-type magnetic balance in a temperature range of 300-800 K.

Electronic Structure Calculation Results:

Mn-Al system \rightarrow Mn and Al atoms are situated in alternating planes spaced at a distance of $c/2$ [1].

Mn atoms situated in adjacent planes ($Mn^{1+}Mn^{2+}$ pairs) \rightarrow AFM coupling.

In order to decrease the weight of the AFM interactions \rightarrow Ni or Ti substitutions for Mn in $Mn_{50}Al_{46}$.

Smaller Ni or Ti moments \rightarrow increase of the total magnetic moment for $Mn_{50}Al_{46}M_4$ ($M = Ni, Ti$).

\rightarrow Smaller total energy for a mixed 1a and 2e Ni occupancy than for an exclusive 2e Ni occupancy.

\rightarrow A lower total energy was found for the alloy with the AFM coupled $Mn^{1+}Mn^{2+}$ pair.

Calculated magnetic moments for the τ phase of $Mn_{50}Al_{46}M_4$ ($M = Ni, Ti$), $Mn_{50}Al_{46}$ and $Mn_{50}Al_{46}Ni_4$ alloys.

	Mn		Al		M		Total	
	$m(\mu_B)$							
$Mn_{50}Al_{46}$	2.41	0.04	-0.08	-	-	-	4.68	0.08
	Mn^{2+}	2.43	0.007	-	-	-	-	-
$Mn_{50}Al_{46}$ (FM)	Mn^{1+}	2.39	0.04	-0.08	-	-	5.02	0.07
	Mn^{2+}	-3.17	-0.01	-	-	-	-	-
$Mn_{50}Al_{46}$ (AFM)	Mn^{1+}	2.35	0.04	-0.07	-	-	4.38	0.08
	Mn^{2+}	2.34	0.04	-0.08	-0.66	-0.004	4.43	0.07
$Mn_{50}Al_{46}Ti_4$	2.45	0.04	-0.08	-	0.63	0.06	4.85	0.08

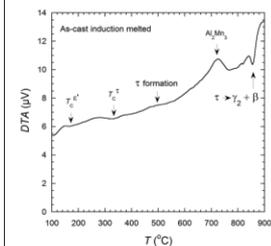
Experimental Results:

During the formation of the τ phase, the following transformation occurs [1,2]:

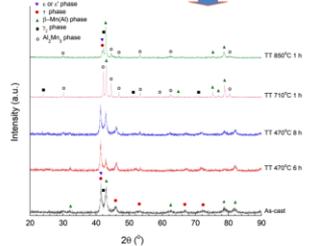
ϵ (disordered hexagonal) \rightarrow ϵ' (ordered orthorhombic) \rightarrow τ (tetragonal)

The ϵ' phase is an ordered ϵ phase [1,2] \rightarrow XRD peaks of these two phases superimpose.

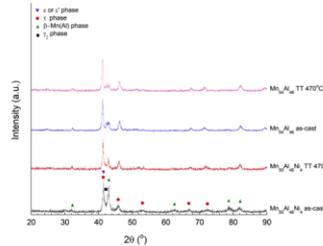
Magnetic measurements can confirm whether the phase is ϵ or ϵ' .



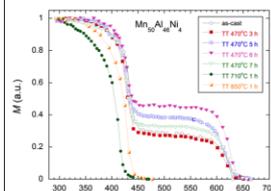
DTA curves for the as-cast $Mn_{50}Al_{46}Ni_4$ alloy.



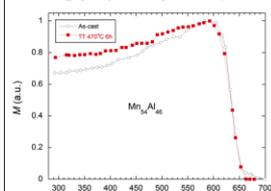
XRD patterns for the $Mn_{50}Al_{46}Ni_4$ alloy after annealing at different temperatures.



XRD patterns for the as-cast and annealed $Mn_{50}Al_{46}$ and $Mn_{50}Al_{46}Ni_4$ alloys.



Temperature dependence of the normalized magnetization of the $Mn_{50}Al_{46}Ni_4$ alloy after annealing at different temperatures.



Temperature dependence of the normalized magnetization of the as-cast $Mn_{50}Al_{46}$ alloy and the sample annealed at 470 °C for 6 h.

As-cast + TT 470 °C 6h

Two magnetic phases:

τ (high T_C) and ϵ' (low T_C).

TT 710 °C + 850 °C

single magnetic phase

behavior (lower T_C)

$Mn_{50}Al_{46}$ samples

single magnetic phase

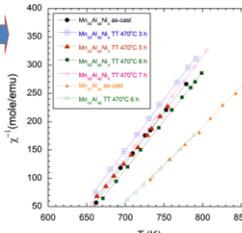
behavior (τ phase).

The experimental data fits the Curie-Weiss law $\chi = C/(T-\theta)$ with the parameters given in the table below.

The average experimental Mn moment in the ordered state was calculated taking into account the concentrations c_1 and c_2 of the ϵ' and τ phases and the theoretical values of Mn moments in the two phases ($\mu_{\epsilon'} = 1.68 \mu_B/\text{Mn atom}$ and $\mu_{\tau} = 2.37 \mu_B/\text{Mn atom}$ [1]) according to the relation: $\mu_{exp} = C_1 \mu_{\epsilon'} + C_2 \mu_{\tau}$.

The μ_{exp} values, the Mn spins S, the ϵ' phase content c_1 , the average Mn moments in the ordered state μ_{exp} and μ_{iso} , the paramagnetic Curie temperatures θ , and the Curie temperatures T_C of the τ phase of the $Mn_{50}Al_{46}Ni_4$ and $Mn_{50}Al_{46}$ alloys.

Sample	μ_{exp} (μ _B)	μ_{iso} (μ _B)	S	c_1 (%)	μ_{τ} (μ _B)	θ (K)	T_C (K)
$Mn_{50}Al_{46}Ni_4$ as-cast	2.11	2.98	1.07	35	2.14	2.05	629
$Mn_{50}Al_{46}Ni_4$ TT 470 °C 3 h	2.10	2.97	1.07	33	2.14	2.04	617
$Mn_{50}Al_{46}Ni_4$ TT 470 °C 5 h	2.15	3.04	1.1	43	2.2	2.09	624
$Mn_{50}Al_{46}Ni_4$ TT 470 °C 6 h	2.15	3.04	1.1	50	2.2	2.13	633
$Mn_{50}Al_{46}Ni_4$ TT 470 °C 7 h	2.10	2.97	1.07	38	2.14	2.07	625
$Mn_{50}Al_{46}$ as-cast	2.40	3.27	1.21	100	2.42	2.37	662
$Mn_{50}Al_{46}$ TT 470 °C 6 h	2.47	3.36	1.25	100	2.50	2.37	655



Reciprocal susceptibility versus temperature curves of the $Mn_{50}Al_{46}Ni_4$ and $Mn_{50}Al_{46}$ alloys. The lines represent the Curie-Weiss fit of the experimental data.

Close values of the μ_{exp} and μ_{iso} values confirm the existence of the ϵ' and τ phases in the investigated $Mn_{50}Al_{46}Ni_4$ alloy.

Conclusions:

- The largest calculated magnetic moment was found for the $Mn_{50}Al_{46}Ni_4$ alloy.
- A lower total energy was found for the alloy with the AFM coupled $Mn^{1+}Mn^{2+}$ pair.
- All of the samples are phase mixtures and are ferromagnetic at room temperature. The τ phase was found along with the ϵ' phase only in the as-cast $Mn_{50}Al_{46}Ni_4$ sample and the ones annealed at 470 °C with a maximum τ phase concentration of 50%.
- Close values of the μ_{exp} and μ_{iso} values confirm the existence of the ϵ' and τ phases in the investigated $Mn_{50}Al_{46}Ni_4$ alloy.
- Ongoing experimental work is currently being conducted to obtain the τ phase as a majority phase in these compounds.

Outline

- Introduction
- MnBi magnetic phase
- MnAl magnetic phase
- **Nanocomposites magnets=Spring magnets**
- Conclusions

a magnetit magnet (1750)

A ferrite magnet (1940)

rare earth based magnet (1980)

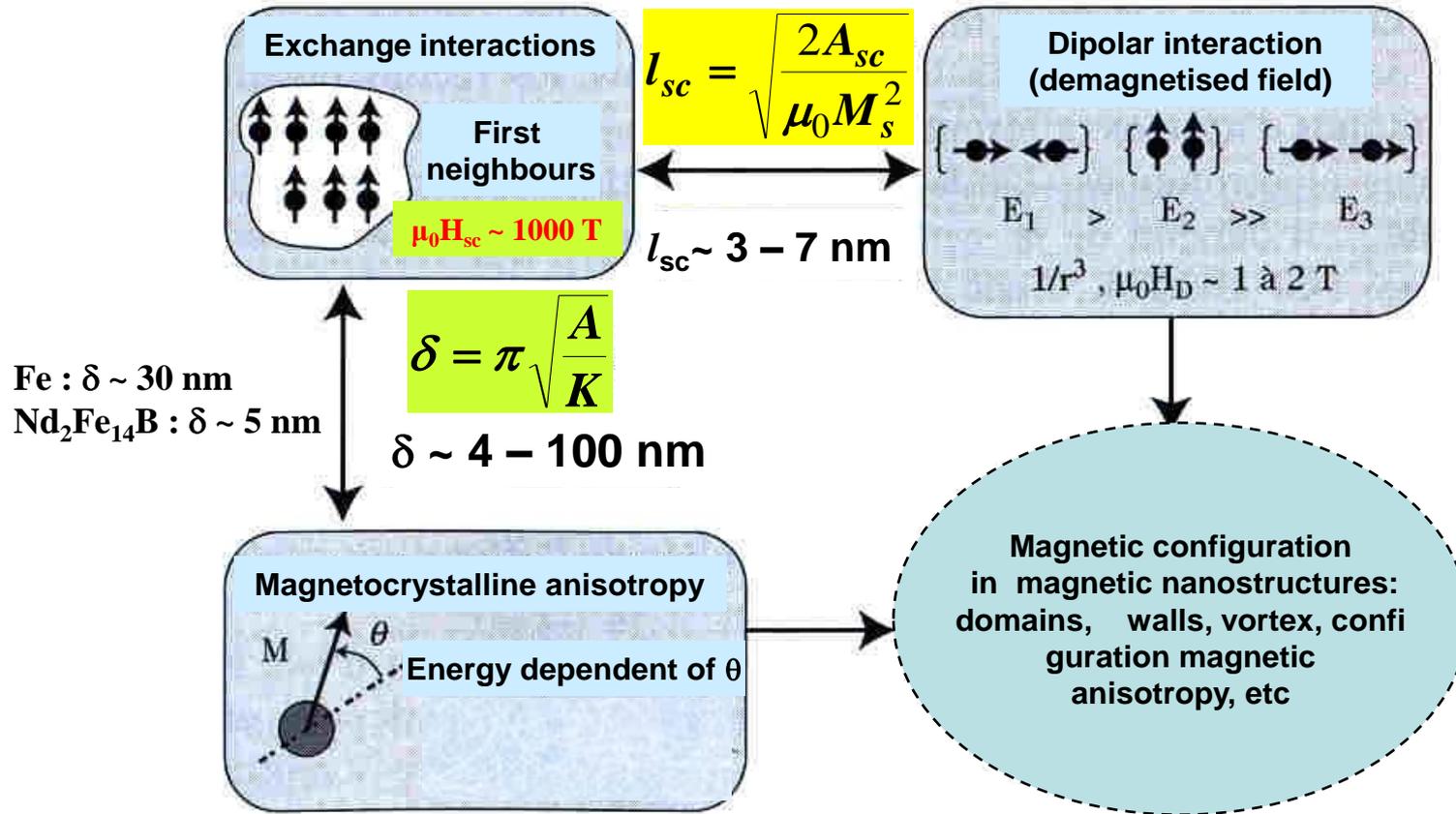
exchange-spring magnets
(20??)



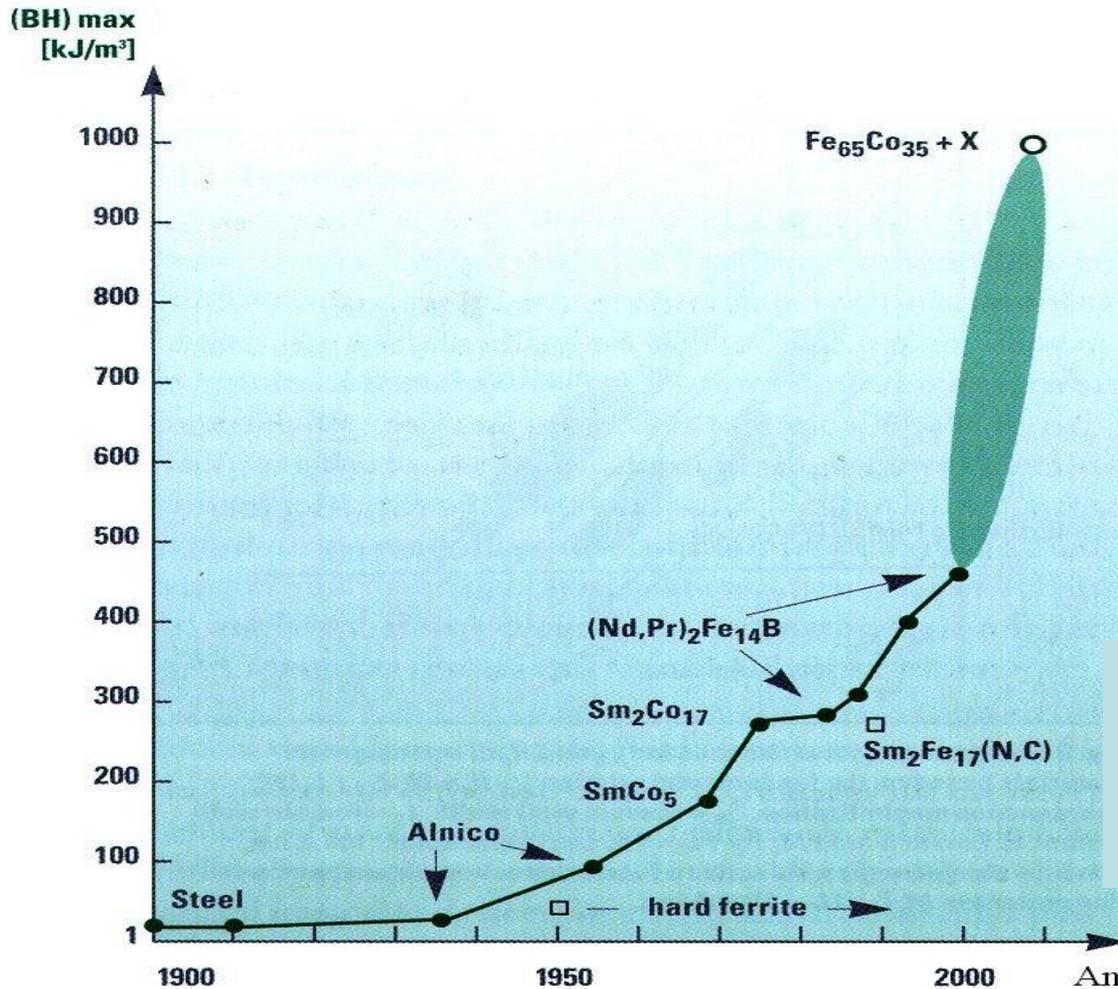
10 cm

All this magnets have the same energy !

Nanophased materials behave differently from their macroscopic counterparts because **their characteristic sizes are smaller than the characteristic length scales** of physical phenomena occurring in bulk materials.



Theoretical predictions:



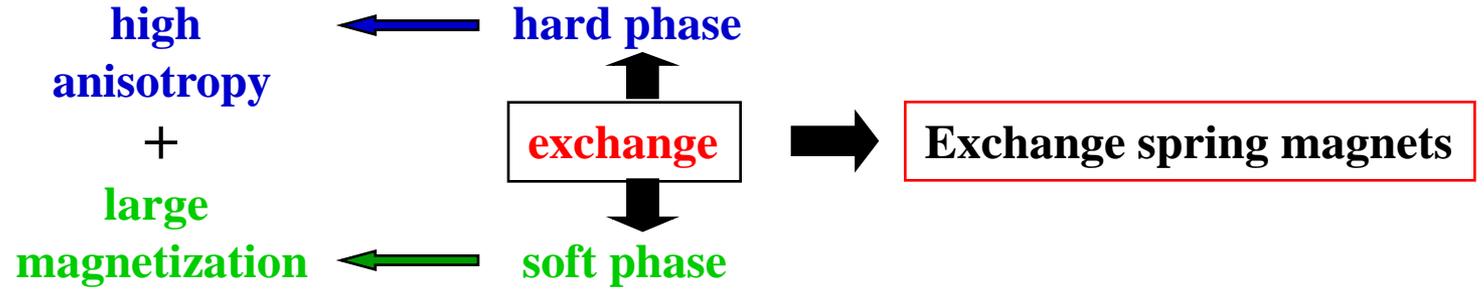
Best magnets on the market:
(BH)_{max} ≈ 500 kJ/m³

(BH)_{max} = 1090 kJ/m³ for
nanostructured multilayers
Sm₂Fe₁₇N₃/Fe₆₅Co₃₅

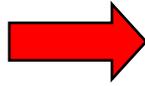
R. Skomski, J. Appl. Phys. 76 (1994) 7059

Kronmuller & Coey *Magnetic Materials*, in
*European White book
on Fundamental Research
in Materials Science*
Max Planck Inst. Metallforschung,
Stuttgart, 2001, 92-96

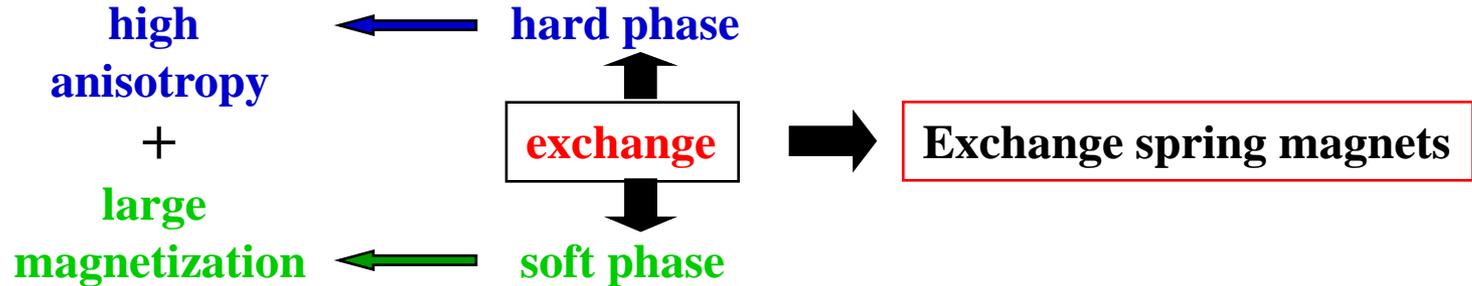
Experimental realisations: ???????????



**Structure
Microstructure**



Soft-hard exchange hardness



$$D_{cr} \approx 2\delta_h$$

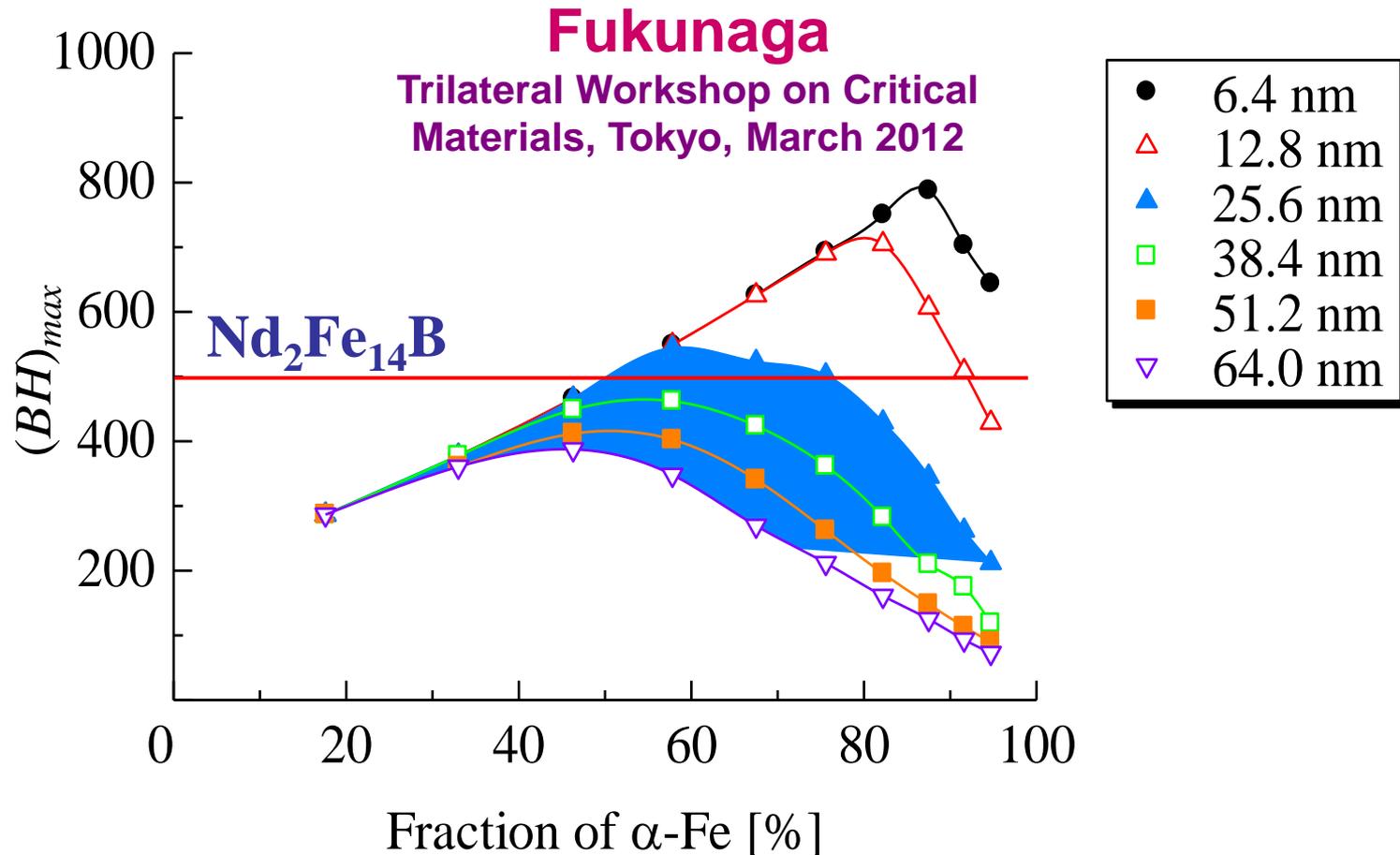
$$\delta_h = \pi \sqrt{A_h / K_h}$$

D_{cr} = soft phase critical dimension

δ_h = width of domain wall in the hard phase

A_h and K_h are the exchange and anisotropy constants

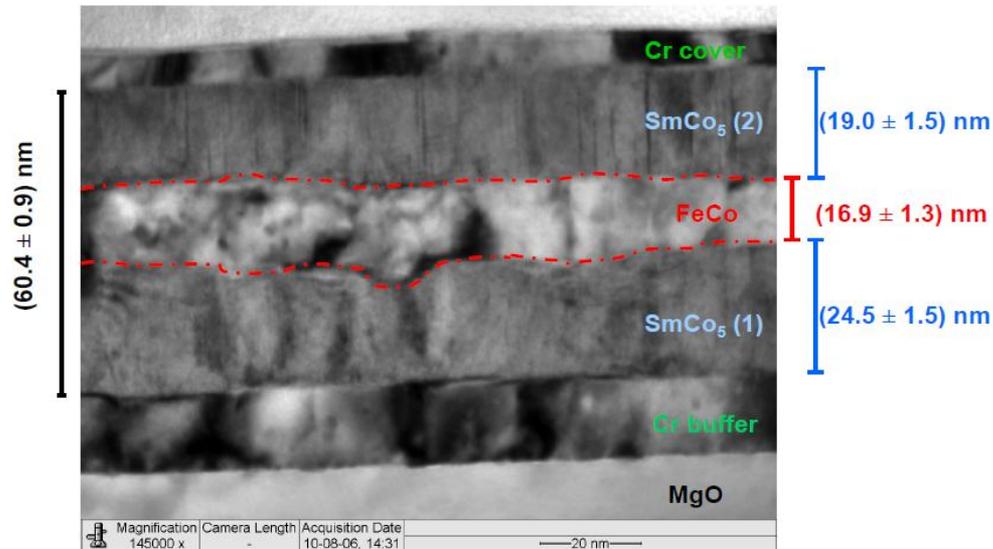
SmCo₅/α-Fe Core-Shell Nanocomposite Magnet



✧ Fukunaga predicted a drastic increase in $(BH)_{\max}$ of SmCo₅/α-Fe as a function of α-Fe fraction. The Dresden Group (Neu et al) obtained similar results in SmCo₅/Fe multilayers (Intermag 2012). Very recently Hono's Group fabricated Fe/Nd-Fe-B multilayers with $(BH)_m=61$ MGOe (Advanced Materials, 2012).

cross-sectional TEM:

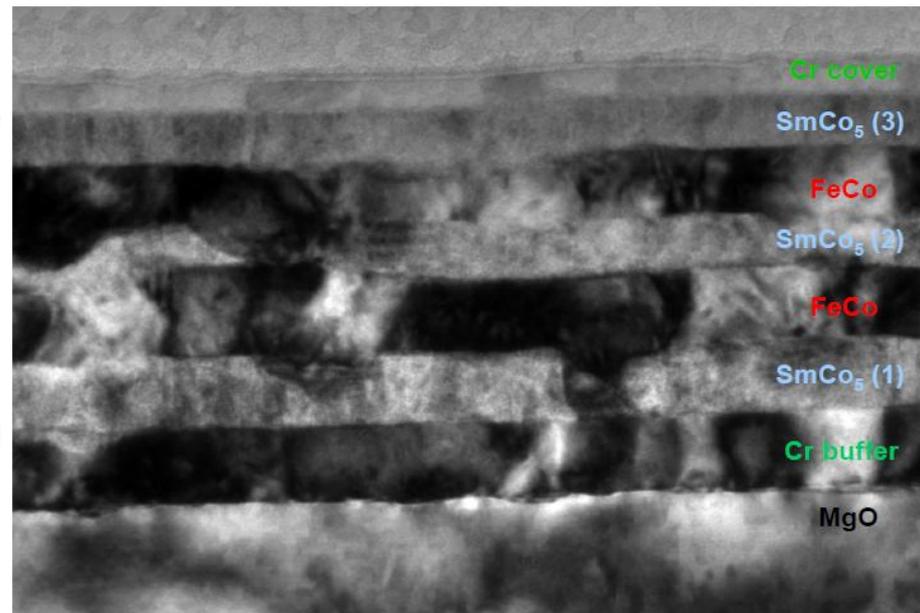
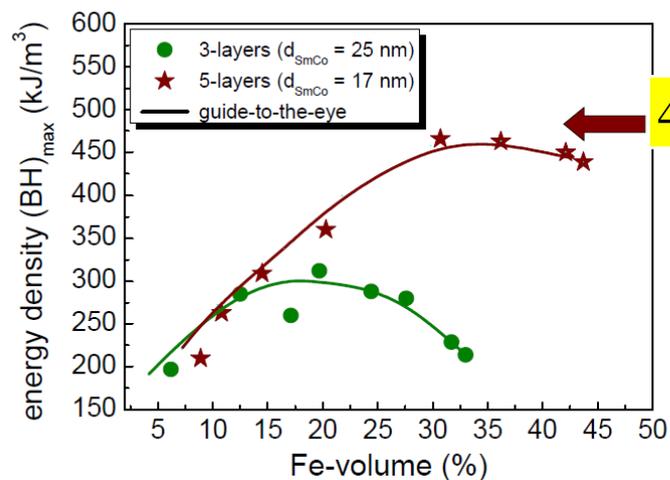
$d_{nom}(Fe) = 12 \text{ nm}$



Ch. Damr

cross-sectional TEM:

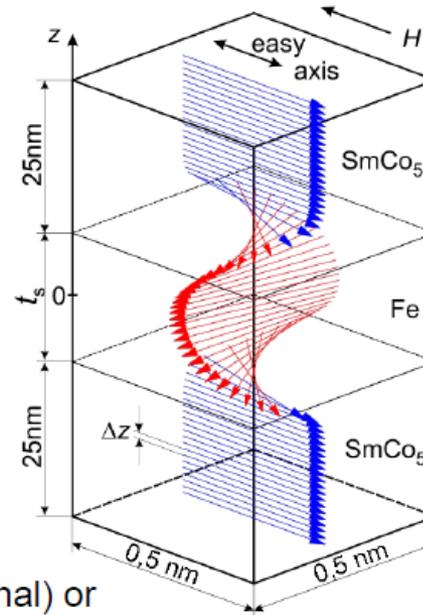
5 layers



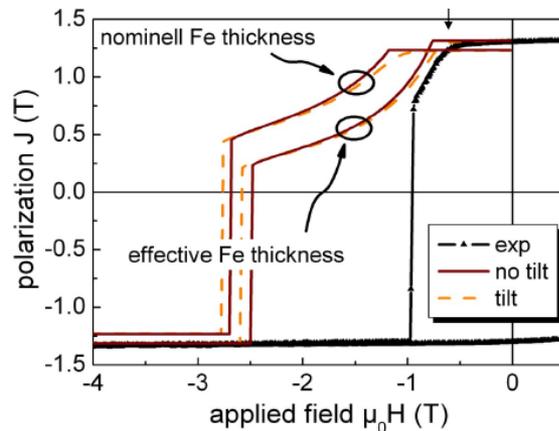
- ❖ code: OOMMF, version 1.2
- ❖ 1-dim approach; no x-y discretization
- ❖ discretization in z-direction: $\Delta z = 0.5 \text{ nm}$
- ❖ intrinsic parameters:

	A (pJ/m)	J_s (T)	K_1 (MJ/m ³)
SmCo ₅	12	1.0	10
Fe	28	2.15	-1.84

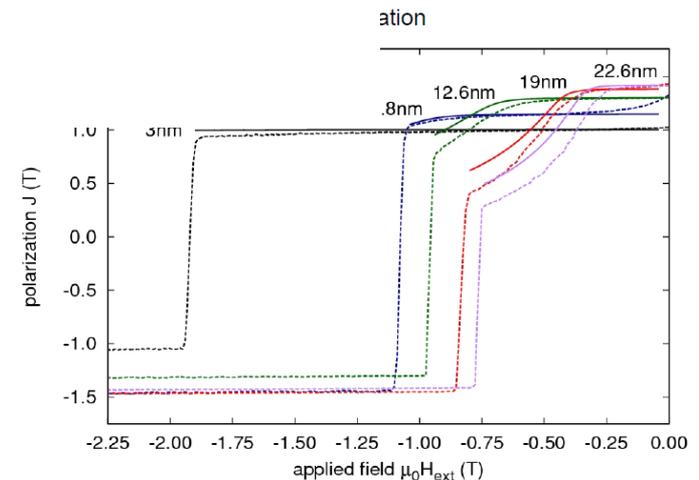
- ❖ strong coupling at interface: $A_i = 20 \text{ pJ/m}$
- ❖ geometry: 2 x 25 nm SmCo₅ + t_s nm Fe (nominal) or reduced SmCo₅ + increased t_s (diffusion)
- ❖ field axis: parallel e.a. or tilted (texture)



11



13



Effect of different surfactants on the formation and morphology of SmCo₅ nanoflakes*

✓ oleylamine (OY),
 ✓ trioctylamine (TOA)
 ✓ oleic acid (OA),

Flake thickness and length, intensity ratio I_{002}/I_{111} values and average grain sizes of the hard SmCo₅ phase.

Original powder	Balls	Surfactants	Milling time (h)	Flake thickness (nm)	Flake length (μm)	I_{002}/I_{111}	Average grain size (nm)
SmCo ₅ ingot after crushed and ground	Mixture of different diameters: Φ4–12 mm	30 wt.% OY	3	20–210	0.5–13	2.5	21
			4	15–170	0.5–11	2.4	16
			5	8–80	0.5–10	1.5	13
		30 wt.% OA	6	8–80	0.2–8	0.6	10
			3	20–210	0.5–13	2.6	21
			4	15–170	0.5–11	2.6	21
		30 wt.% TOA	5	8–80	0.5–10	1.8	20
			6	8–80	0.2–8	0.8	15
			0	–	1–40 ^a	–	–
		100 wt.% TOA	5	–	0.5–20 ^a	2	11
			10	–	0.5–20 ^a	–	–
			2.5	80–300	0.5–10	3.4	21
		40 wt.% TOA	5	30–150	0.3–8	2.5	11
			0.25	–	1–17 ^a	22.8	–
			0.5	500–1600	1–17	19.5	–
			1	80–800	1–15	7.2	–
			2	80–420	1–12	4.0	–
			2.5	80–200	1–10	3.8	21
			3	60–180	0.5–10	2.0	21
4	50–150		0.5–10	1.5	18		
5	30–120		0.5–8	0.8	11		
Jet-milled SmCo ₅	Single diameter: Φ4 mm	5 wt.% OA	0	–	0.5–10 ^a	63.6	–
			0.25	–	0.3–10 ^a	43.1	–
			0.5	–	0.3–10 ^a	22.2	–
		2 wt.% OA	1	20–350	0.5–10	16.5	–
			2	10–200	0.4–10	4.3	–
			5	8–150	0.3–10	1.4	9
			5	10–210	0.3–10	0.4	9
			5	8–150	0.3–10	1.4	9
		5 wt.% OA	5	8–150	0.3–10	1.4	9
			5	8–150	0.3–10	1.4	9
		10 wt.% OA	5	5–100	0.3–10	2.0	9

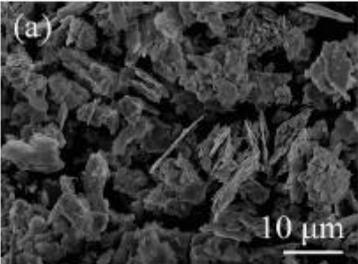
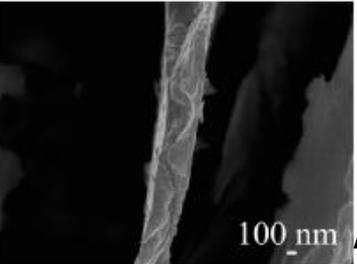
^a Particle size.

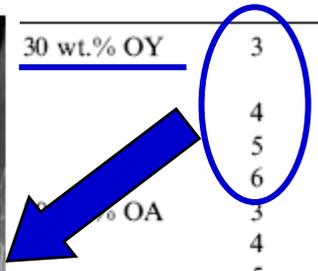
* Liyun Zheng, Baozhi Cui, George C. Hadjipanayis, Acta Materialia 59 (2011) 6772–6782

Effect of different surfactants on the formation and morphology of SmCo_5 nanoflakes*

- ✓ oleylamine (OY),
- ✓ trioctylamine (TOA)
- ✓ oleic acid (OA),

Flake thickness and length, intensity ratio I_{002}/I_{111} values and average grain sizes of the hard SmCo_5 phase.

Original powder	Balls	Surfactants	Milling time (h)	Flake thickness (nm)	Flake length (μm)	I_{002}/I_{111}	Average grain size (nm)	
		30 wt.% OY	3	20-210	0.5-13	2.5	21	
			4	15-170	0.5-11	2.4	16	
			5	8-80	0.5-10	1.5	13	
			6	8-80	0.2-8	0.6	10	
			30 wt.% OA	3	20-210	0.5-13	2.6	21
				4	15-170	0.5-11	2.6	21
		5		8-80	0.5-10	1.8	20	
		6		8-80	0.2-8	0.8	15	
		30 wt.% TOA		0	-	1-40 ^a	-	-
				5	-	0.5-20 ^a	2	11
			10	-	0.5-20 ^a	-	-	
		100 wt.% TOA	2.5	80-300	0.5-10	3.4	21	
5	30-150		0.3-8	2.5	11			
40 wt.% TOA	0.25	-	1-17 ^a	22.8	-			
	0.5	500-1600	1-17	19.5	-			
	1	80-800	1-15	7.2	-			
	2	80-420	1-12	4.0	-			
	2.5	80-200	1-10	3.8	21			
	3	60-180	0.5-10	2.0	21			
	4	50-150	0.5-10	1.5	18			
	5	30-120	0.5-8	0.8	11			
5 wt.% OA	0	-	0.5-10 ^a	63.6	-			
	0.25	-	0.3-10 ^a	43.1	-			
	0.5	-	0.3-10 ^a	22.2	-			
	1	20-350	0.5-10	16.5	-			
	2	10-200	0.4-10	4.3	-			
2 wt.% OA	5	8-150	0.3-10	1.4	9			
	5 wt.% OA	5	10-210	0.3-10	0.4	9		
		5	8-150	0.3-10	1.4	9		
	10 wt.% OA	5	5-100	0.3-10	2.0	9		



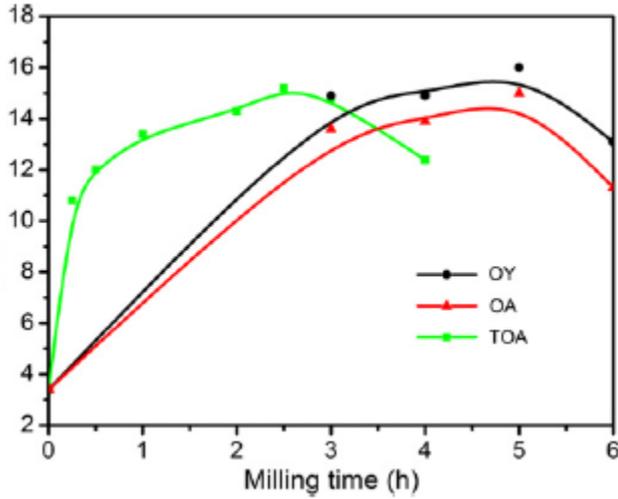
micro-size powders became flakes. ↑

Effect of different surfactants on the formation and morphology of SmCo_5 nanoflakes*

- ✓ oleylamine (OY),
- ✓ trioctylamine (TOA)
- ✓ oleic acid (OA),

Flake thickness and length, intensity ratio I_{002}/I_{111} values and average grain sizes of the hard SmCo_5 phase.

Original powder	Balls	Surfactants	Milling time (h)	Flake thickness (nm)	Flake length (μm)	I_{002}/I_{111}	Average grain size (nm)		
SmCo ₅ ingot after crushed and ground	Mixture of different diameters: $\Phi 4$ –12 mm	30 wt.% OY	3	20–210	0.5–13	2.5	21		
			4	15–170	0.5–11	2.4	16		
			5	8–80	0.5–10	1.5	13		
		30 wt.% OA	6	8–80	0.2–8	0.6	10		
			3	20–210	0.5–13	2.6	21		
			4	15–170	0.5–11	2.6	21		
		30 wt.% TOA	5	8–80	0.5–10	1.8	20		
			6	8–80	0.2–8	0.8	15		
			5	–	1–40 ^a	–	–		
		100 wt.% TOA	5	–	0.5–20 ^a	2	11		
			10	–	0.5–20 ^a	–	–		
			2.5	80–300	0.5–10	3.4	21		
		Jet-milled SmCo ₅	Single diameter: $\Phi 4$ mm	5 wt.% OA	5	30–150	0.3–8	2.5	11
					0.25	–	1–17 ^a	22.8	–
					0.5	500–1600	1–17	19.5	–
				2 wt.% OA	1	80–800	1–15	7.2	–
					2	80–420	1–12	4.0	–
					2.5	80–200	1–10	3.8	21
3	60–180				0.5–10	2.0	21		
4	50–150				0.5–10	1.5	18		
5	30–120				0.5–8	0.8	11		
5	–				0.5–10 ^a	63.6	–		
0.25	–	0.3–10 ^a	43.1	–					
0.5	–	0.3–10 ^a	22.2	–					
5	20–350	0.5–10	16.5	–					
5	10–200	0.4–10	4.3	–					
5	8–150	0.3–10	1.4	9					
5	10–210	0.3–10	0.4	9					
5	8–150	0.3–10	1.4	9					
5	5–100	0.3–10	2.0	9					



^a Particle size.

Our researches

- hard magnetic phases of **SmCo₅, SmCo₃Cu₂, R₂Fe₁₄B**
- soft magnetic phases of **α-Fe, Fe-Co (~20 or 10 wt%)**

SmCo₅	large anisotropy
SmCo₃Cu₂	large coercivity
R₂Fe₁₄B	best magnets

Our researches

- hard magnetic phases of **SmCo₅, SmCo₃Cu₂, R₂Fe₁₄B**
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Magnetic Hard/Soft nanocomposites – Spring magnets

(SmCo₅, SmCo₃Cu₂, R₂Fe₁₄B)+x% (α-Fe or Fe₆₅Co₃₅)

- **composition, x= 10 or 22 wt % Fe**
- **milling time**
- **conventional annealing/short time annealing**

Material preparation

- milling of the powders in a high energy planetary mill
- heat treatments (temperatures and duration)

Starting materials :

- **hard magnetic phases**
ingots – prepared by melting
- Fe NC 100.24 powder (Höganäs), (< 40 μm) and
- $\text{Fe}_{65}\text{Co}_{35}$ obtained by melting



Mechanical milling experiments:

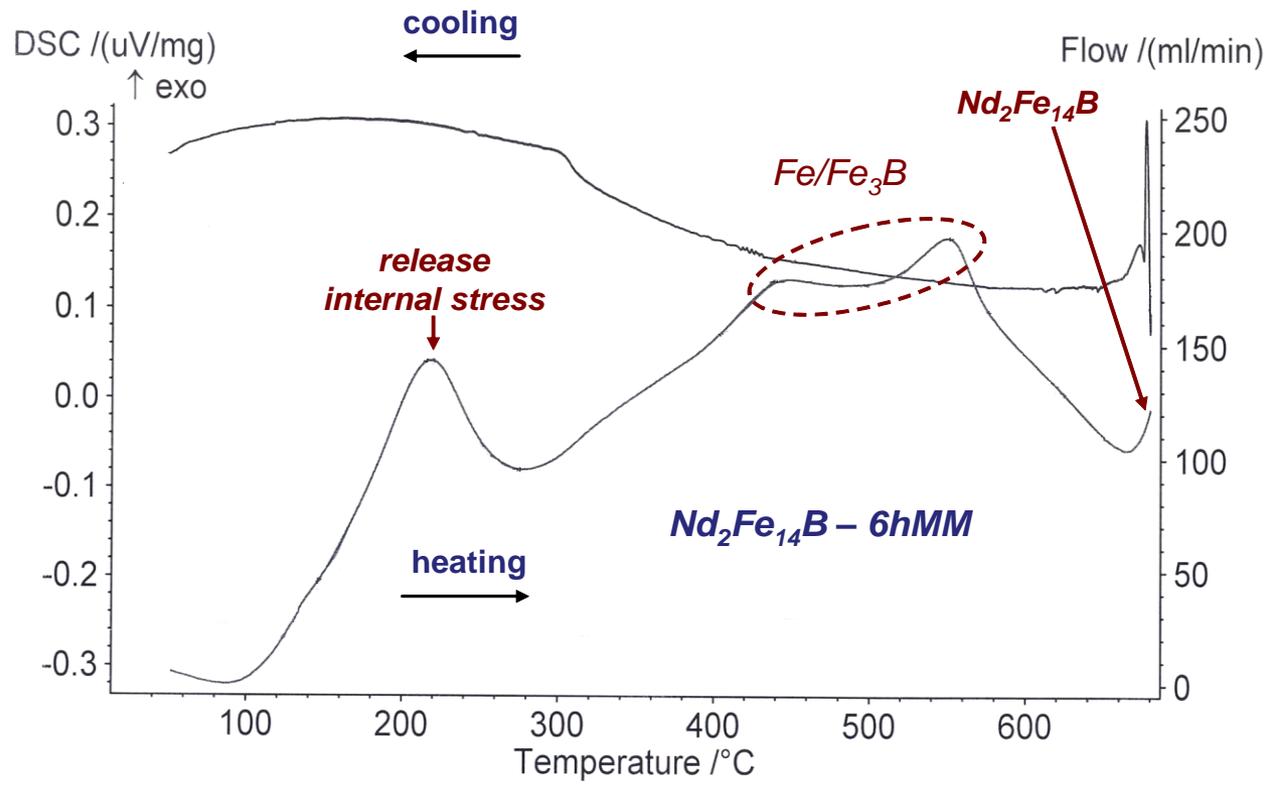
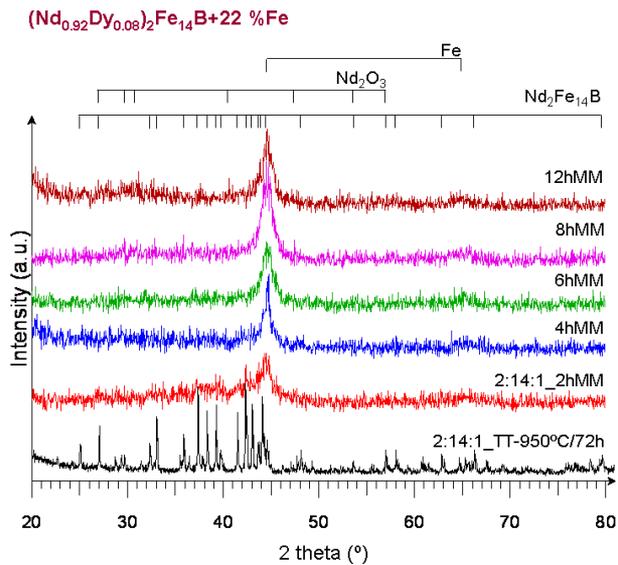
- premilling of hard and soft magnetic ingots
- **hard magnetic** + $\alpha\text{-Fe}$ (or $\text{Fe}_{65}\text{Co}_{35}$) mixed powders – milled in Ar for **1.5 – 12 h**

Annealing:

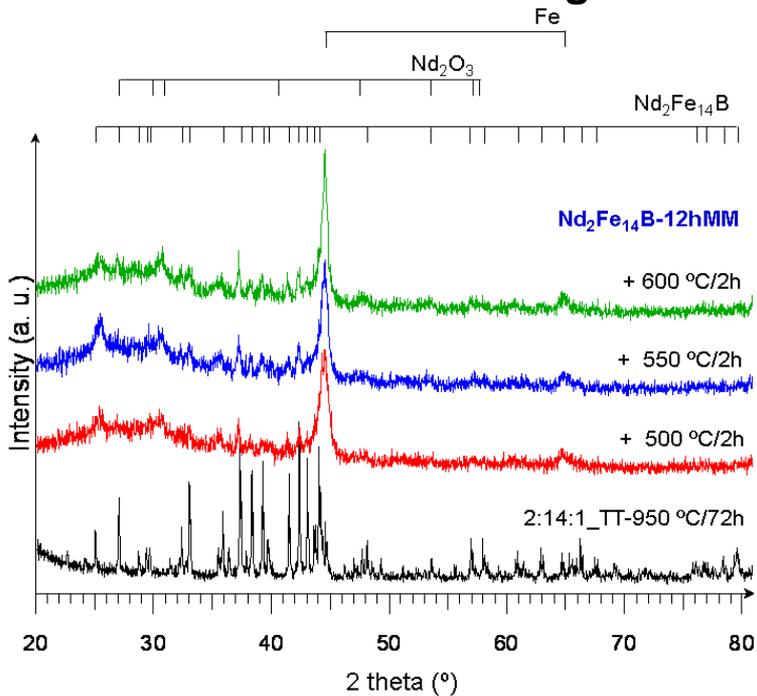
- conventional annealing: in vacuum/450-650 ° C for 0.5 up to 10 h.
- short time annealing: in argon/700, 750 or 800 ° C for 0.5 to 3 min.

Material characterisation

- **X-rays diffraction (XRD)**
- **DSC measurements**
- **Electron microscopy (SEM and TEM)**
 - morphology**
 - chemical composition checked by EDX**
- **Magnetic measurements**

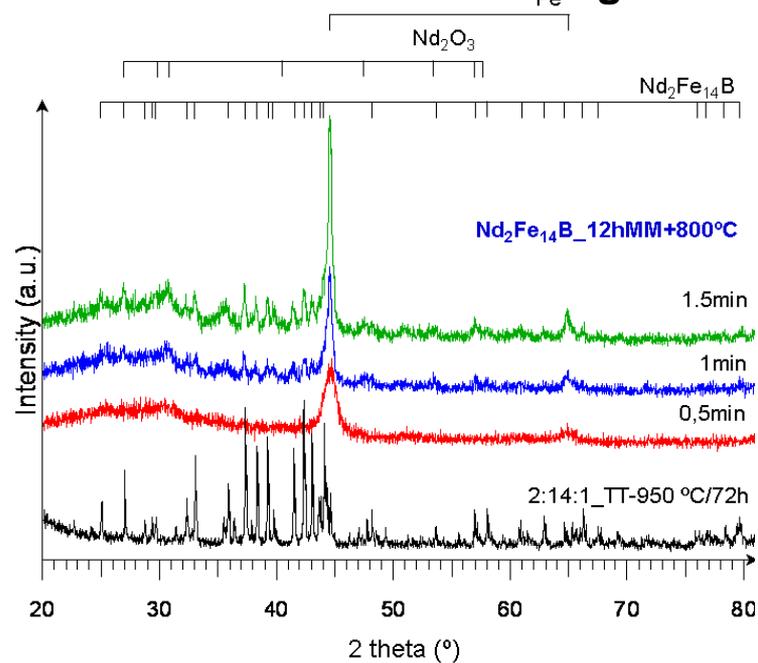


Classical annealing

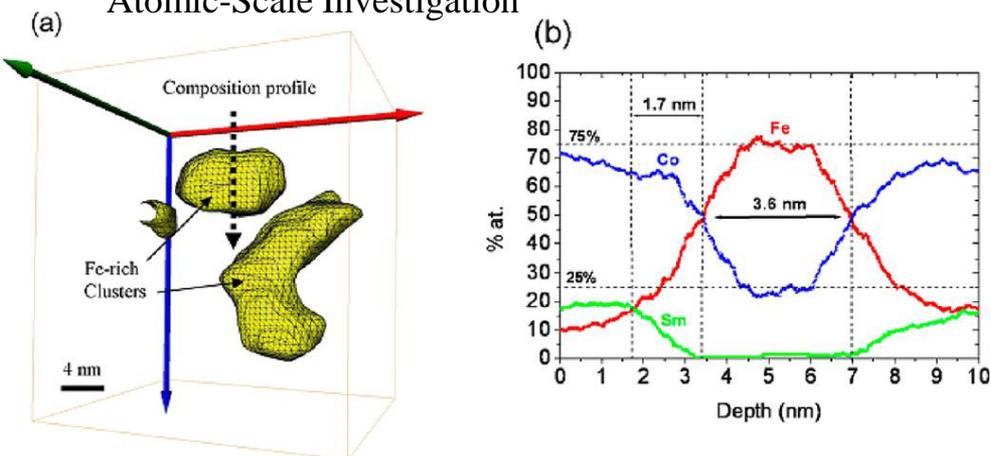


S. Gutoiu, V. Pop et al., J. Optoelectron. Adv. Mater. 12 (2010) 2126-2131

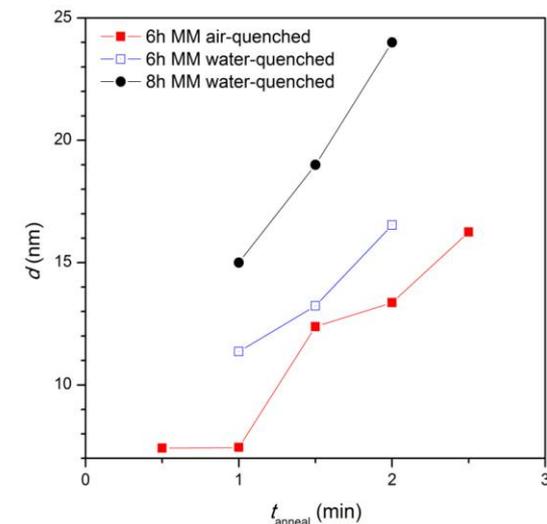
Short time annealing

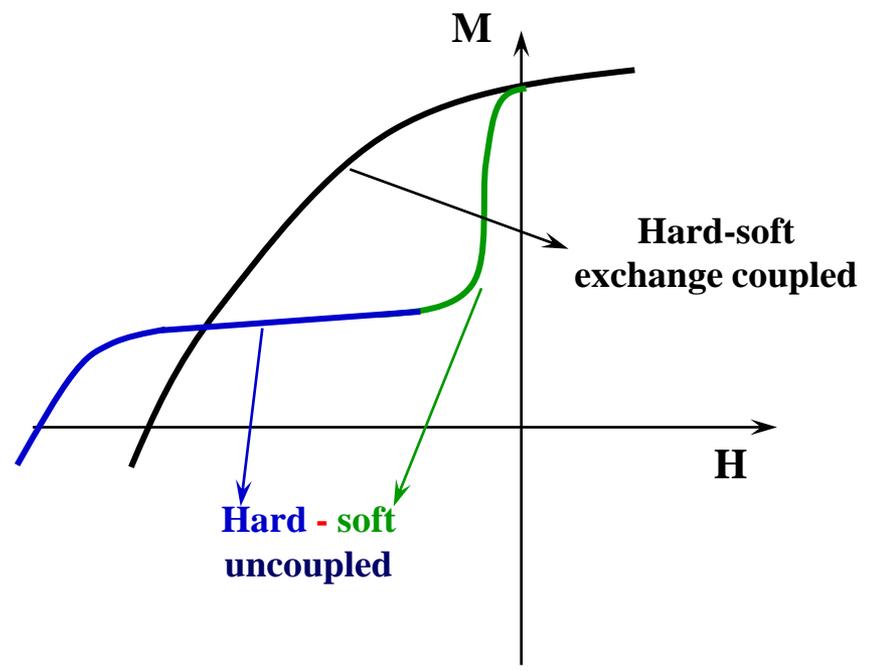
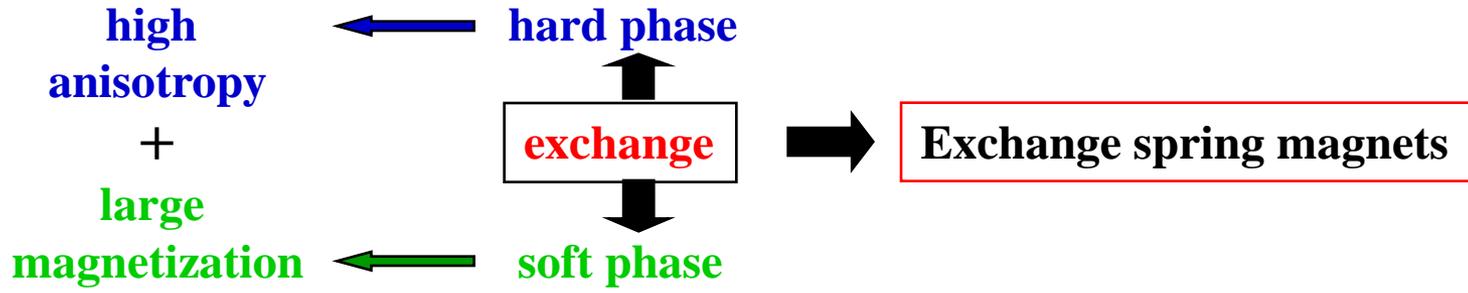


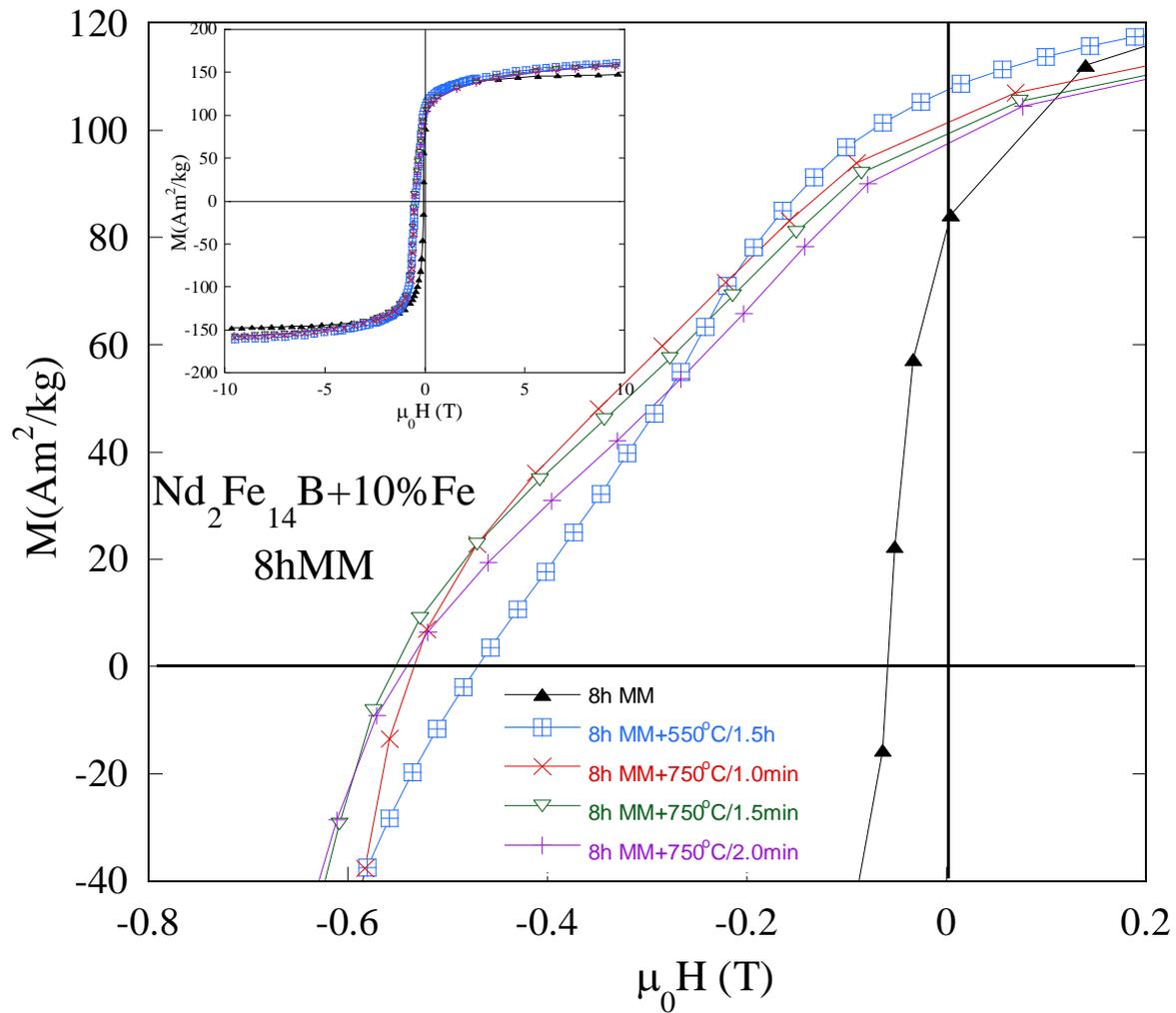
Atomic-Scale Investigation



R. Lardé, J-M. Le Breton, A. Maître, D. Ledue, O. Isnard, V. Pop and I. Chicinaş, J. Phys. Chem., 117 (2013) 7801







$\text{Nd}_2\text{Fe}_{14}\text{B} + 10\% \alpha\text{-Fe}; 8\text{h MM}$

Classical annealing

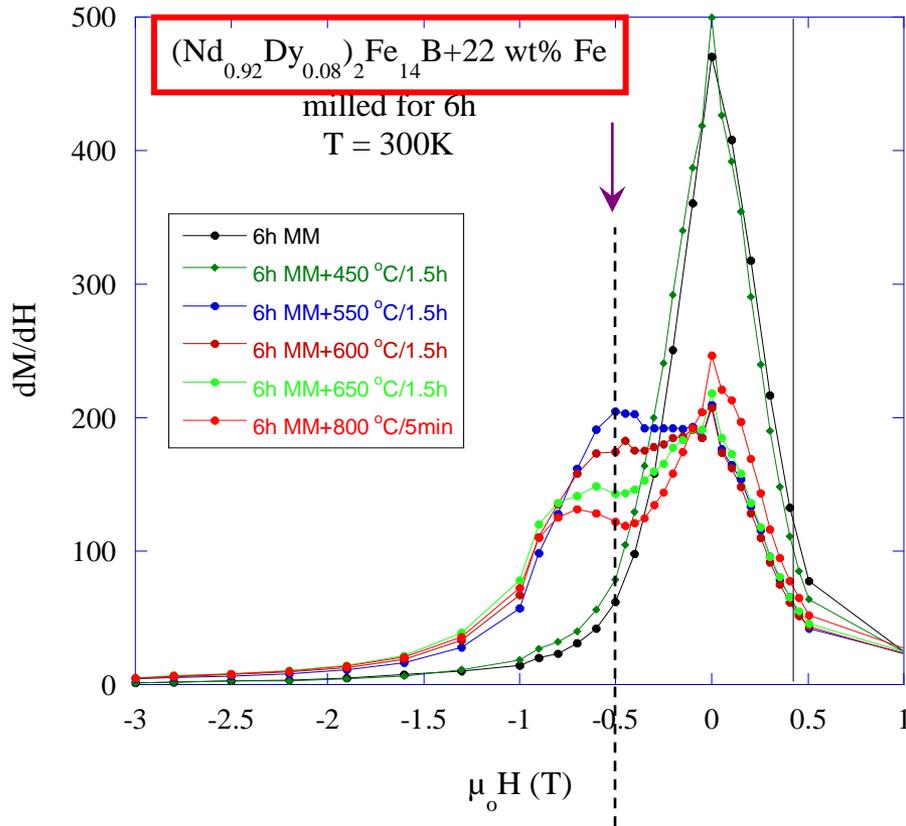


Short time annealing

H_c - evident increasing

M_r - small decreasing

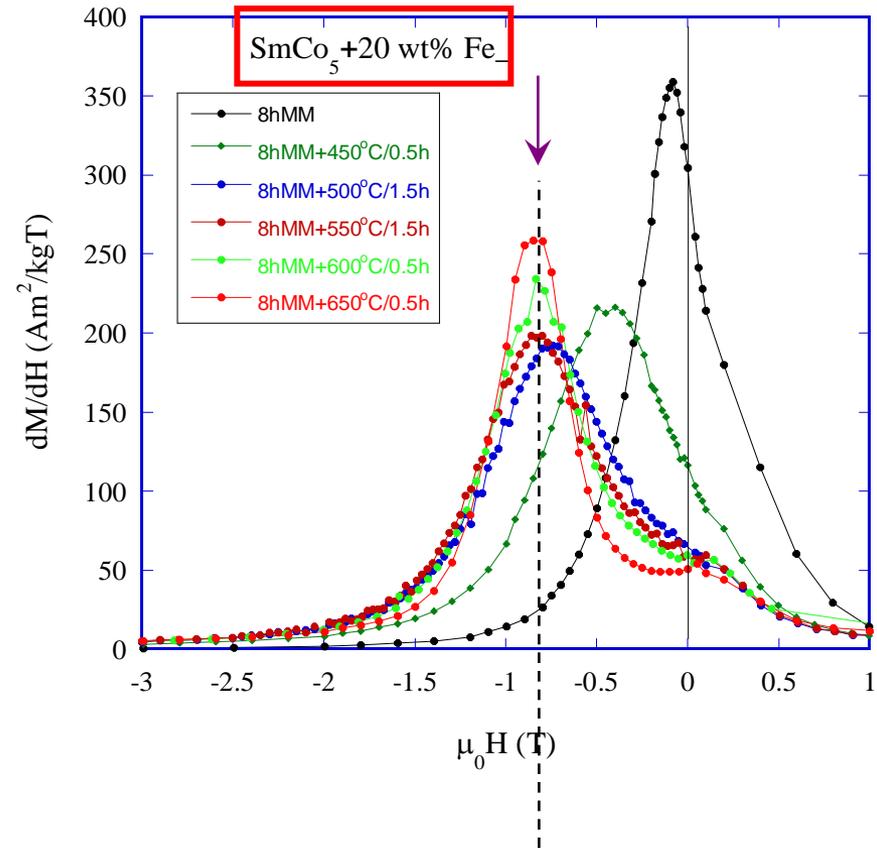
The influence of the type of hard magnetic phase



$\text{R}_2\text{Fe}_{14}\text{B}/\text{Fe}$

Lower coercivity

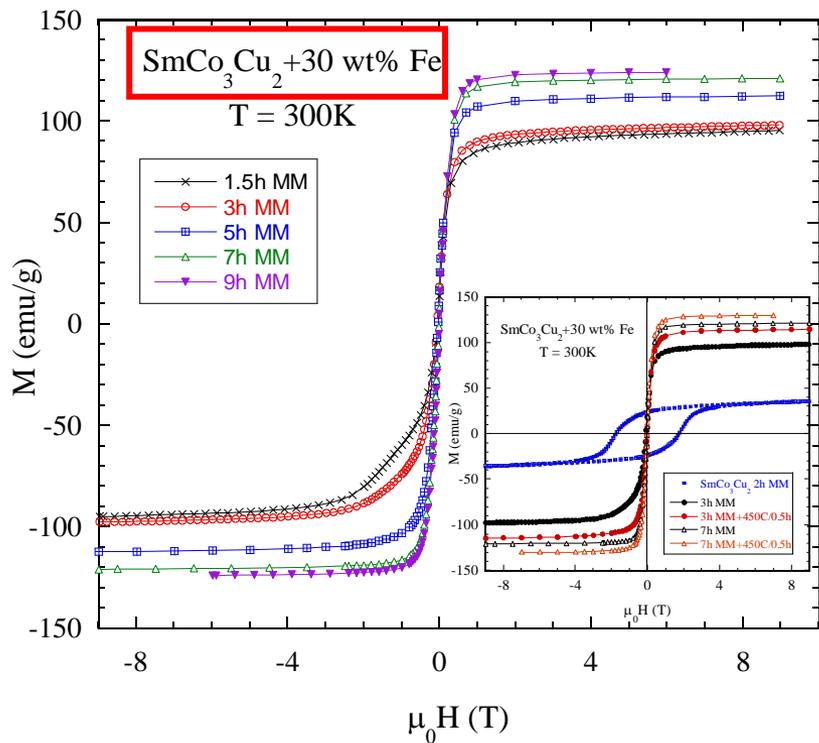
Two peaks,
poor hard/soft magnetic coupling



SmCo_5/Fe

Higher coercivity

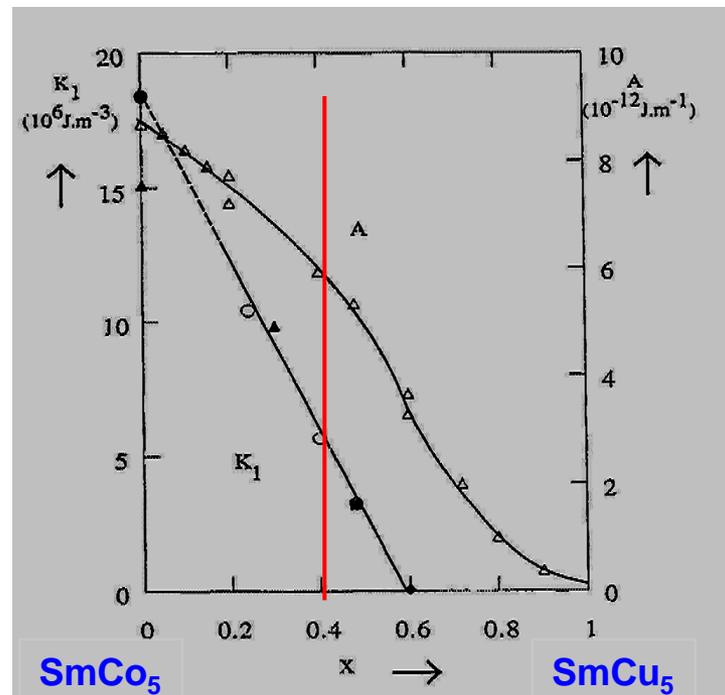
One peak,
good hard/soft magnetic coupling



the importance of the
intrinsic anisotropy*

*D. Givord, O. Isnard, V. Pop, I. Chichinas, JMMM 316 (2007)

This behavior was connected with coercivity mechanism of the SmCo_3Cu_2 phase given by the microstructure [1-2] and diminishing of the intrinsic coercivity by Co substitution with Cu [3].



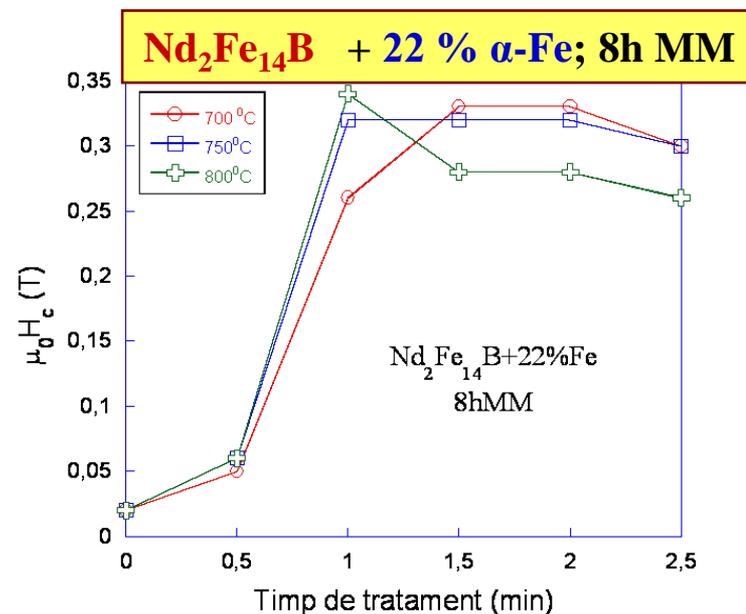
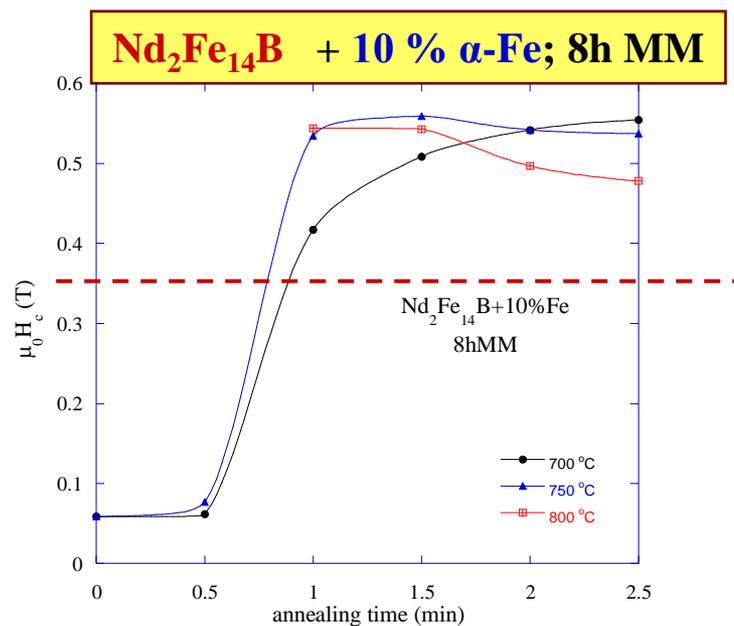
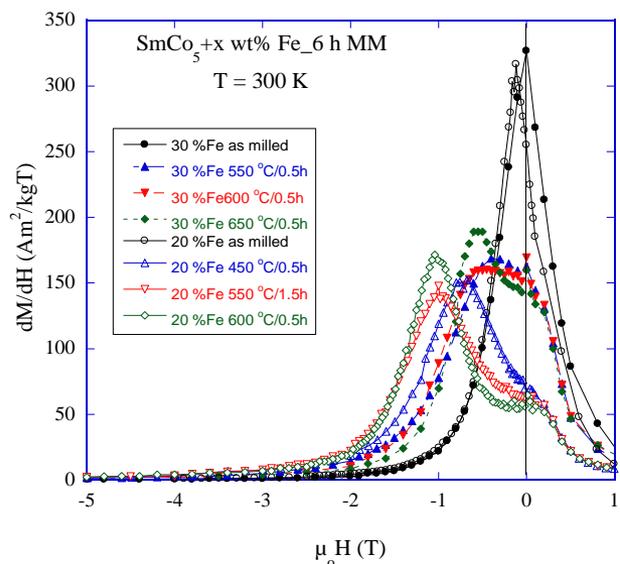
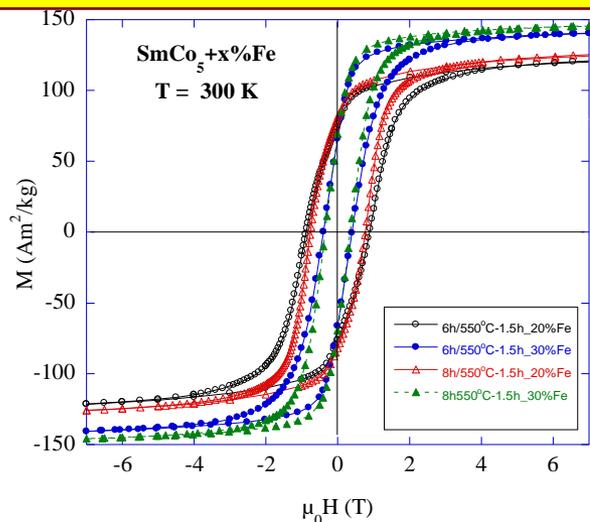
- [1] E. Estevez-Rams, J. Fidler, A. Penton, J.C. Tellez-Blanco, R.S. Turtelli, R. Grossinger, J. Alloys Compounds, 283 (1999) 327.
- [2] P. Kersch, A. Handstein, K. Khlopov, O. Gutfleisch, D. Eckert, K. Nenkov, J.-C. Tellez-Blanco, R. Grossinger, K.-H. Müller, L. Schultz, J. Magn. Magn. Matter. 290–291 (2005) 420.
- [3] E. Lectard, C.H. Allibert, R. Ballou, J. Appl. Phys. 75 (1994) 6277.



The influence of the hard/soft ratio

SmCo₅+x% Fe (x=20 or 30),

milled 6 and 8 h and annealed at 550° C for 1.5 h*



* V. Pop, O. Isnard, D. Givord, I. Chicinas, JMMM 310 (2007) 2489

V. Pop, O. Isnard, D. Givord, I. Chicinas, J. M. Le Breton, JOAM 8 (2006) 494

V. Pop et al., J. Alloys Compd. (2011)

V. Pop et al., J. Alloys Compd. 581 (2013) 821–827



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² Université Grenoble Alpes, Institut Néel, Grenoble, F-38042 France

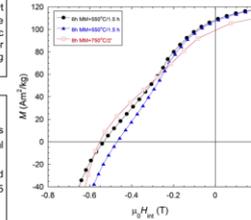
³ CNRS, Institut Néel, Grenoble, F-38042 France

⁴ Materials Science and Engineering Department, Technical University of Cluj-Napoca, RO-400641 Romania

Abstract: This study presents the effect of different milling times and short time annealing on the structural and magnetic properties of Nd₂Fe₁₄B/10wt % Fe nanocomposites prepared by high energy ball milling. The XRD peaks of the hard magnetic phase disappear after milling due to the damaging of the Nd₂Fe₁₄B crystal structure. After annealing, the characteristic peaks of the hard magnetic phase are restored with a limited growth of the soft magnetic phase crystallites. The magnetic behavior was investigated from hysteresis curves and dM/dH vs. H plots. The best exchange coupling was obtained for the 6 h milled sample annealed at 700 °C for 2 minutes with a maximum coercive field value of 0.44 T. Taking into account the milling and annealing conditions, the Nd₂Fe₁₄B/α-Fe exchange coupling is analyzed.

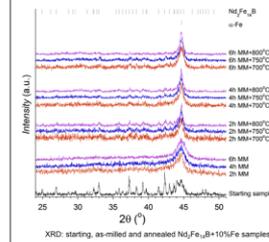
Experimental:

- The Nd₂Fe₁₄B hard phase was prepared by induction melting in an Ar atmosphere, followed by annealing in vacuum at 950 °C for 68 h. The ingot was ground to a fine powder under 500 μm. The soft magnetic phase (12 g of NC 100.24 commercial Fe powder – Höganas product) was milled with 5 ml benzene for 4 h in an inert Ar atmosphere with a ball to powder weight ratio of 10:1.
- The Nd₂Fe₁₄B powder was mixed with the pre-milled Fe phase in a weight ratio of 90% Nd₂Fe₁₄B/10% Fe. The mixture was dry-milled in Ar for 2, 4 and 6 h respectively with a ball to powder weight ratio of 10:1. The milled samples were annealed in an Ar atmosphere at 700, 750 and 800 °C for 0.5-2.5 min and quenched in water.
- X-ray diffraction measurements were performed on a Bruker D8 Advance diffractometer using Cu K_α radiation.
- Magnetic measurements were carried out on powder samples fixed in epoxy resin using the extraction method at 300 K in applied fields up to ±10T. Assuming isolated spherical magnetic particles we used a demagnetization factor of 1/3 for magnetic data.



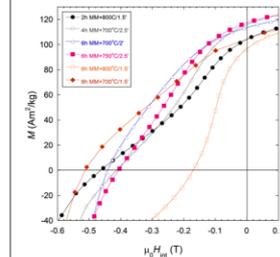
For classically annealed samples, the interphase exchange coupling is better after 6 h MM compared to 8 h MM [1,2]. What will happen after short time annealing?

Results and Discussions:



Damaging of the crystal structure and increased internal stress and defect density by MM.
Hard phase XRD peaks disappear even after 2 h MM.
The Nd₂Fe₁₄B crystal structure is restored after annealing.

$d = 10-25$ nm
No excessive growth of the soft phase crystallites during the short time annealing.
The crystallite sizes of the annealed 6 h MM samples are smaller than those of the annealed 8 h MM samples [1].

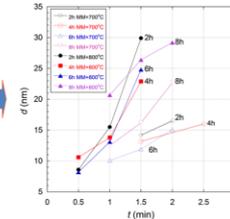


The crystallite sizes were determined using the Scherrer formula for the peak with $2\theta = 82.3^\circ$:

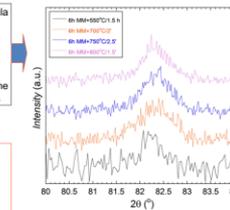
$$d = \frac{K\lambda}{\beta \cos \theta}$$

d is the average crystallite size, K is the shape factor, λ is the wavelength, β is the FWHM of the peak and θ is the Bragg angle.

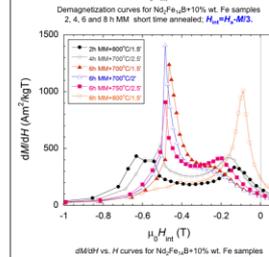
The coercivity of the short time annealed 6 h MM samples increases with annealing time.
Progressive restoration of the Nd₂Fe₁₄B crystal structure.
Coercivity decreases with increasing annealing temperature → larger α-Fe crystallites.



Mean crystallite sizes of α-Fe vs. annealing time for the as-milled and annealed samples. The lines are guides for the eye.

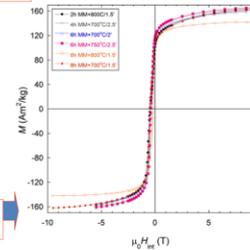


XRD peaks centered at $2\theta = 82.3^\circ$ corresponding to α-Fe.



Low field peaks → badly coupled/uncoupled α-Fe
Higher field peaks → exchange-coupled nanocomposite.

Saturation is not reached up to 10 T.



±10T magnetization curves for Nd₂Fe₁₄B/10wt % Fe. 2, 4, 6 and 8 h MM samples short time annealed at 700-800 °C; H₀=H₀M₀.

Progressive decoupling with increasing annealing time/temperature.
Increasingly larger soft phase crystallites and possible phase alterations.

Conclusions:
 > Short time annealing restores the structure of the hard phase destroyed by milling with a limited growth of the soft magnetic crystallites.
 > The exchange coupling strength increases with milling time possibly due to smaller soft phase crystallites and a better homogeneity of the mixture.
 > The best exchange coupling was obtained for the 6 h MM sample annealed at 700 °C for 2 minutes with a maximum coercive field value of 0.44 T.
 > The coercivity of the short time annealed 6 h MM samples is slightly lower than previously reported values on the classically annealed 6 h and short time annealed 8 h MM samples, however, they show a higher remanence.
 > The diminishing of the coercivity could be attributed to the pre-milling of the soft phase with benzene.

[1] V. Pop, S. Gaitan, E. Dorci, O. Isnard and I. Chicinaș, J. Alloys Compd., 599, 999 (2011).
 [2] V. Pop, S. Gaitan, E. Dorci, C. Leventon, O. Isnard, I. Chicinaș and O. Isnard, J. Alloy Compd., 581, 821 (2013).
 [3] S. Gaitan, E. Dorci, O. Isnard, I. Chicinaș, and V. Pop, J. Optoelectron. Adv. Mater., 12, 2126 (2010).

Acknowledgment
 This work was supported by the Romanian Ministry of Education and Research, Grant No. PN-II-D-PCE-2012-4-0470.

Conclusions

- **MnBi LTP**: large coercivity at high temperature \Rightarrow a good candidate for performance spring magnets
- **MnAl**: stabilisation of τ with conservation of Mn moments
- The *structure and microstructure* \Rightarrow strong impact on hard/soft exchange hardness.
- **Intrinsic anisotropy** \Rightarrow the **strength** of the interphase exchange coupling
- *Annealing* linked to the *recrystallisation temperature* of soft phases and hard magnetic phases; *recover* the crystallinity of the hard phase and *hinder the increase* of Fe crystallites.
- For *higher α -Fe concentration* the magnetic properties are poorer because non correlation with Fe size crystallites.
- The *short time annealing* is more appropriate for higher coercivity of the nanocomposites.

Thank you for your attention

**This contribution was supported by the
Romanian Ministry of Education and Research, grant PN-II-ID-PCE-2012-4-0470**